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decimetres per mole  $\text{dm}^3 \text{ mol}^{-1}$  for gases and cubic centimetres per mole  $\text{cm}^3 \text{ mol}^{-1}$  for liquids and solids

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web zero degrees celsius or 273 K is defined as standard temperature and one atmosphere or 760 mm Hg is defined as standard pressure the volume occupied by one mole of a gas is called molar volume therefore the molar volume of a gas at standard temperature and pressure STP is  $22.4 \text{ L}$

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volume of gas moles  $V_m$  rearranging this moles of gas  $\frac{V}{n}$

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