

Flame Test Lab

Name Lani Simmons Period 8th Date 10/27

Objective: To recognize that the color of light emitted by a heated element is unique and can be translated into energy that is coming from the movement of electrons within the atoms.

Background: The natural position of an electron in an energy level is called its *ground state*. When an atom absorbs energy, it is in its *excited state*. This happens when electrons are heated in a flame or electrocuted. These excited electrons jump to higher energy levels. Eventually, the excited electrons fall back down to their ground state, releasing their extra energy as a packet of light called a *photon*.

The color of light corresponds to a specific amount of energy. This color varies from one substance to another, because the location of electrons within each substance is different. Just as we have unique fingerprints, each element has a unique color of light (amount of energy) that it releases. As a result, the colors we observe when a substance is heated may be used to identify the element.

Pre-lab Questions:

1. Fill in the blanks with the terms "excited, ground, or light."

When an atom absorbs energy, the electrons move from their ground state to their excited state. Then, the electrons release energy, as they fall back down from their excited state to their ground state. This release of energy is in the form of light.

2. What is a photon? A photon is a tiny particle that comprises waves of electromagnetic radiation. As shown by Maxwell, photons are just electric fields traveling through space. Photons have no charge, no resting mass, and travel at the speed of light.

Data: Hint: Each color will only be used once in the data table below.

Metal Salt	Circle the Color of the Flame that You Observed:					
Calcium Chloride	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)
Copper Chloride	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)
Lithium Chloride	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)
Sodium Chloride	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)
Barium Chloride	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)
Strontium Chloride	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)
Unknown	Yellow/Green	Yellow	Blue-green	Orange	Red	Crimson (Darker Red)

Unknown Number _____

Post-lab Questions:

1. In this lab, you tested six different metal salts. Each metal salt contained one metal atom and one chlorine atom. How do you know that the color observed for each salt came from exciting the *metal* atom and not the *chlorine* atom?

2. Identify your unknown metal atom in the space below. *CHLORIDE* should not be part of your answer!

Unknown Number	Identification

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Active Learning Patricia Ann Mabrouk,2007 This symposium series book focuses on the application of active learning methods in teaching analytical science broadly defined at both the undergraduate and graduate levels The volume includes a wide range of examples of how these methods are being applied at public and private community colleges four year colleges and graduate research universities in the United States and abroad Flash Gordon (1965) Gold Key Comics Alex Raymond,2014-08-15 In the company of Dale Arden Flash Gordon embarked for the planet Mongo in 1934 That was in the Sunday funnies in a page drawn by Alex Raymond and written anonymously by former pulp fiction editor Don Moore This space opera became one of King Features Syndicate s most popular features and Raymond s illustrative art was to have a strong influence on many of the young artists who began drawing for comic books in the late 1930s and the early 1940s Tom Hickey Sheldon Moldoff Jack Lehti George Papp Mac Raboy Dan Barry etc Flash Gordon entered comic books early in 1936 by way of reprints in King Comics His battles with the merciless Ming a sort of galactic Fu Manchu unfolded in the magazine from the first issue In the early 1940s Dell began issuing now and then Flash Gordon reprint titles Later in the decade came an occasional comic book offering Flash adventures especially written and drawn for this magazine The artist was Paul Norris who also began drawing the Jungle Jim newspaper page in 1948 Harvey Publications tried reprinting the Raymond material in 1950 and 1951 giving up after a few issues King Features experimented with publishing comic books in the late 1960s These used original material and the Flash Gordon book made use of such artists as Al Williamson a devoted Raymond disciple Gil Kane and Reed Crandall When King quit Charlton took over and finally Gold Key The final Whitman Flash Gordon comic book was printed in 1982 He reappeared briefly in 1987 as part of a team that included Mandrake and the Phantom in the TV inspired Defenders of the Earth Energy Research Abstracts ,1991 *Spring Meeting* American Geophysical Union. Meeting,1991 **Metals Abstracts** ,1983-07 **Government Reports Announcements & Index** ,1995 **Late Night Labs to Accompany Beran Lab Manual 8E** Beran,2008-09-09 This collection of challenging experiments will help get readers up to speed on laboratory techniques safety and experimental procedures Each experiment is presented with concise objectives a comprehensive list of techniques and detailed lab intros and step by step procedures Beran also integrates numerous Dry Labs experiments that do not involve chemicals but reinforce the reader s knowledge of general chemistry topics such as nomenclature oxidation numbers and atomic and molecular structure *Physical Science Lab Manual Answer Key* ,2006-02-23 Help students explore and understand the world around them With the full color Physical Science text students learn the properties of matter elements compounds electricity and sound and light Students reading significantly below grade level gain practice in working with data and sharpen their abilities to infer classify and theorize Lexile Level 840 Reading Level 3 4 Interest Level 6 12 **Chemistry Test Key Units 1-10 (RES)** Responsive Education Solutions Staff,2011-08-01 Key Individual Test Key for Chemistry Units 1 10

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