

Exercises

1. Select appropriate answers for the following.
 - i. Which of the following has highest electron gain enthalpy ?
A. Fluorine B. Chlorine
C. Bromine D. Iodine
 - ii. Hydrides of group 16 are weakly acidic. The correct order of acidity is
A. $\text{H}_2\text{O} > \text{H}_2\text{S} > \text{H}_2\text{Se} > \text{H}_2\text{Te}$
B. $\text{H}_2\text{Te} > \text{H}_2\text{O} > \text{H}_2\text{S} > \text{H}_2\text{Se}$
C. $\text{H}_2\text{Te} > \text{H}_2\text{Se} > \text{H}_2\text{S} > \text{H}_2\text{O}$
D. $\text{H}_2\text{Te} > \text{H}_2\text{Se} > \text{H}_2\text{O} > \text{H}_2\text{S}$
 - iii. Which of the following element does not show oxidation state of +4 ?
A. O B. S C. Se D. Te
 - iv. HI acid when heated with conc. H_2SO_4 forms
A. HIO_3 B. KIO_3
C. I_2 D. KI
 - v. Ozone layer is depleted by
A. NO B. NO_2
C. NO_3 D. N_2O_5
 - vi. Which of the following occurs in liquid state at room temperature ?
A. HIO_3 B. HBr
C. HCl D. HF
 - vii. In pyrosulfurous acid oxidation state of sulfur is
A. Only +2 B. Only +4
C. +2 and +6 D. Only +6
 - viii. Stability of interhalogen compounds follows the order
A. $\text{BrF} > \text{IBr} > \text{ICl} > \text{ClF} > \text{BrCl}$
B. $\text{IBr} > \text{BrF} > \text{ICl} > \text{ClF} > \text{BrCl}$
C. $\text{ClF} > \text{ICl} > \text{IBr} > \text{BrCl} > \text{BrF}$
D. $\text{ICl} > \text{ClF} > \text{BrCl} > \text{IBr} > \text{BrF}$
 - ix. BrCl reacts with water to form
A. HBr B. $\text{Br}_2 + \text{Cl}_2$
C. HOBr D. $\text{HOBr} + \text{HCl}$
 - x. Chlorine reacts with excess of fluorine to form.
A. ClF B. ClF_3
C. ClF_2 D. Cl_2F_3
 - xi. In interhalogen compounds, which of the following halogens is never the central atom.
A. I B. Cl C. Br D. F
 - xii. Which of the following has one lone pair of electrons ?
A. IF_3 B. ICl
C. IF_5 D. ClF_3
 - xiii. In which of the following pairs, molecules are paired with their correct shapes ?
A. $[\text{I}_3]$: bent
B. BrF_3 : trigonal bipyramid
C. ClF_3 : trigonal planar
D. $[\text{BrF}_4]$: square planar
 - xiv. Among the known interhalogen compounds, the maximum number of atoms is
A. 3 B. 6 C. 7 D. 8
2. Answer the following.
 - i. Write the order of the thermal stability of the hydrides of group 16 elements.
 - ii. What is the oxidation state of Te in TeO_3 ?
 - iii. Name two gases which deplete ozone layer.
 - iv. Give two uses of ClO_2 .
 - v. What is the action of bromine on magnesium metal ?
 - vi. Write the names of allotropic forms of selenium.
 - vii. What is the oxidation state of S in H_2SO_5 ?
 - viii. The $\text{p}K_a$ values of HCl is -7.0 and that of HI is -10.0. Which is the stronger acid ?
 - ix. Give one example showing reducing property of ozone.

Maharashtra Oct 2014 Chemistry

Shasha Hu



Maharashtra Oct 2014 Chemistry:

Fundamental Of Fluid Dynamics Dr. S. Kiran, Dr. V. Sumathi, Dr. Raviteja Surakasi, Dr. Cephas A. Vanderhyde, 2022-10-11

The study of fluids and their motion and static behaviour is known as fluid mechanics. An explanation of what we mean by fluid should come first. Fluids are substances that undergo continuous deformation when subjected to shear tangential stress regardless of the magnitude of the applied force. Another way to characterise fluidity is to say that it is incompatible with shear stress in its resting state. Molecules make up fluids. However, the macroscopic or average influence of several molecules is of more importance in engineering applications. The macroscopic influence is what we often see and quantify. This means that we ignore the behaviour of the individual molecules and instead think of the fluid as a continuum or endlessly divisible entity. The study of fluid mechanics encompasses a wide range of disciplines that are difficult to categorise. Scientific studies have identified two distinct types of flows: laminar and turbulent, which researchers use to categorise the degree of order and chaos present in a fluid. Single phase flow and multiphase flow are two distinct phenomena in fluid physics. Since fluids may undergo a phase transition, condensation or evaporation during the flow, changing from the single phase flow to the multi phase flow, the final boundary like all the boundaries in fluid mechanics isn't crisp. In addition, two phase or multimaterial flows may be analysed as if they were single phase flows.

Annual Report on the Development of the Indian Ocean Region (2016) Rong Wang, Cuiping Zhu, 2017-06-01. This book focuses on international relations in the Indian Ocean region and covers many policy aspects of Modi's India. Written by leading scholars of international and Indian Ocean studies, this annual report includes both a strategic review and the major events and related data in this region. It also discusses the origin, aims, frameworks and regional and global impact of India's development under the Modi administration, offering readers a full and authentic picture of the most recent developments in India. This year's Annual Report is the fourth of this kind and the only one to include the Indian Ocean region of China.

QRS for BDS I Year Jyotsna Rao, 2020-06-14. Quick Review Series for BDS 1st Year is an extremely exam oriented book. The book contains a collection of the last 25 years' questions of General Anatomy including Embryology and Histology, Physiology, Biochemistry, Oral Histology and Dental Anatomy in accordance with the BDS 1st year syllabus. The book will serve the requirements of BDS 1st year students to prepare for their examinations and help PG aspirants in quick review of important topics. Unique collection of last 25 years' solved questions asked in major university examinations across India. Simple, well illustrated, lucid in content and style in two color format. Book contains numerous flowcharts and tables for easier understanding. Perfectly segregated into 6 sections: General Anatomy including Embryology and Histology, Physiology, Biochemistry, Oral Histology and Dental Anatomy, Self assessment Questions and Previous Years Question Bank. Self assessment section of this book includes key points to remember, MCQs with answers and viva questions for practical exam preparation. Sample question papers on all the subjects. Thoroughly revised and updated with latest questions from all major universities across India. Addition of new MCQs and viva questions.

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The Enigmatic Realm of **Maharashtra Oct 2014 Chemistry**: Unleashing the Language is Inner Magic

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