# CHEMISTRY PAPER 1 (THEORY)

Maximum Marks: 70

Time Allowed: Three hours

Candidates are allowed additional 15 minutes for only reading the paper.

They must NOT start writing during this time.

This paper is divided into four sections - A. B. C and D.

Answer all questions.

Section A consists of one question having sub-parts of one mark each.

Section B consists of ten questions of two marks each.

Section C consists of seven questions of three marks each, and

Section D consists of three questions of five marks each.

Internal choices have been provided in one question each in Section B, Section C and Section D.

All working, including rough work, should be done on the same sheet as, and adjacent to the rest of the answer.

The intended marks for questions or parts of questions are given in brackets [ ]. Balanced equations must be given wherever possible and diagrams where they are helpful.

When solving numerical problems, all essential working must be shown.

In working out problems, use the following data:

Gas constant R = 1.987 cal deg' mot' = 8.314 JK' mot' = 0.0821 dm' atm K'mot'

I I atm = I dm3 atm = 1013 J. I Faraday = 96500 coulombs.

Avogadro's number =  $6.023 \times 10^{19}$ .

#### SECTION A – 14 MARKS

#### Ouestion 1

(A) Fill in the blanks by choosing the appropriate word(s) from those given in the [4×1] brackets:

[stable, low, aldehyde, unstable, 6, 4, ethane, Clemmensen's, 2, 3, carboxylic acid, high, propane, Rosenmund's]

- (i) The primary alcohols are easily oxidised first into \_\_\_\_\_ and then into
- (ii) The intermediate activated complex in a chemical reaction is highly \_\_\_\_\_ due to \_\_\_\_\_ energy.
- (iii) The coordination number and exidation state of the complex K<sub>4</sub>[Fe(CN)<sub>6</sub>] are \_\_\_\_\_ and \_\_\_\_ respectively.
- (iv) Propanone on reaction with zinc-amalgam in presence of conc. HCl gives and the reaction is known as \_\_\_\_\_ reduction.

This Paper consists of 8 printed pages.

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#### Table of Contents Jnv 2014 Class 11 Chemistry Question Paper

- 1. Understanding the eBook Jnv 2014 Class 11 Chemistry Question Paper
  - The Rise of Digital Reading Jnv 2014 Class 11 Chemistry Question Paper
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Jnv 2014 Class 11 Chemistry Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Jnv 2014 Class 11 Chemistry Question Paper
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Jnv 2014 Class 11 Chemistry Question Paper

- Personalized Recommendations
- Jnv 2014 Class 11 Chemistry Question Paper User Reviews and Ratings
- Jnv 2014 Class 11 Chemistry Question Paper and Bestseller Lists
- 5. Accessing Jnv 2014 Class 11 Chemistry Question Paper Free and Paid eBooks
  - o Jnv 2014 Class 11 Chemistry Question Paper Public Domain eBooks
  - Jnv 2014 Class 11 Chemistry Question Paper eBook Subscription Services
  - o Jnv 2014 Class 11 Chemistry Question Paper Budget-Friendly Options
- 6. Navigating Jnv 2014 Class 11 Chemistry Question Paper eBook Formats
  - o ePub, PDF, MOBI, and More
  - $\circ$  Jnv 2014 Class 11 Chemistry Question Paper Compatibility with Devices
  - Jnv 2014 Class 11 Chemistry Question Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - o Adjustable Fonts and Text Sizes of Jnv 2014 Class 11 Chemistry Question Paper
  - Highlighting and Note-Taking Jnv 2014 Class 11 Chemistry Question Paper
  - o Interactive Elements Jnv 2014 Class 11 Chemistry Question Paper
- 8. Staying Engaged with Jnv 2014 Class 11 Chemistry Question Paper
  - o Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Jnv 2014 Class 11 Chemistry Question Paper
- 9. Balancing eBooks and Physical Books Inv 2014 Class 11 Chemistry Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Inv 2014 Class 11 Chemistry Question Paper
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Jnv 2014 Class 11 Chemistry Question Paper
  - Setting Reading Goals Jnv 2014 Class 11 Chemistry Question Paper
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Jnv 2014 Class 11 Chemistry Question Paper

- Fact-Checking eBook Content of Jnv 2014 Class 11 Chemistry Question Paper
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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