Chapter

7

THERMOCHEMISTRY

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The second second

MULTIPLE CHOICE QUESTIONS

| 1. | Spen | taneous reacti | ions are | er. | | | | | CLHIR 05) | |
|----|--|-------------------|----------|----------------|----------------------|--------|----------------------------|---------------------|----------------------|--|
| | (40) | Reversible | (b) | Irreversible | (c) | Mon I | rneversible | (4) | None of these | |
| 2. | The standard heat changes occur at: (BWP 08, GRW 06) | | | | | | | | | |
| | (a) | 25°C and 2 at | | | | (b) | 298 K and 1 a | | | |
| | (c) 25°C and 1 mm hg | | | | | (d) | 273 K and 1 at | mo. | | |
| 3. | H" + OH"→ H ₂ O the change in enthalpy for reaction is called: (UHROS) | | | | | | | | | |
| | (a) | Heat of reacti | | | | (b) | heat of forms | | | |
| | (c) | Heat of neutri | | | | (4) | Heat of comb | | | |
| 4. | | | | y in a chemica | al react | ion is | | | es place directly or | |
| | | ectly. It is call | | | | | | | 1, GRW 07, BWP 10) | |
| | (40) | Henry's law | (b) | Charlie's law | . Allen and a second | (c) | | (d) | Graham's law | |
| 5. | The enthalpy change when one mole of substance is completely burnt in excess of oxygen is | | | | | | | | | |
| | called: (LHR 07) | | | | | | | | | |
| | (a) Enthalpy of atomization | | | | | (P) | Enthalpy of neutralization | | | |
| | (c) Enthalpy of Combustion | | | | | (d) | Enthalpy of formation | | | |
| 6. | Enthalpy change for the reaction: $CH_{4(n)} + 2O_{2(n)} \rightarrow CO_{2(n)} + 2H_2O_{(0)}$ is called enthalpy of: $(LH_2O_{(0)})$ | | | | | | | | | |
| | (a) | Formation | (b) | Combustion | | (c) | Neutralization | (d) | Atomization | |
| 7. | Standard enthalpies are measured at: (ESO 11, LHR 09) | | | | | | | | | |
| | (a) | 273 K | (b) | 2:9BK | | (<) | 373 K | (4) | All of these | |
| 8. | | | | | | | | IK 10, GRW, LHR 11) | | |
| | (a) | Evaporation | (b) | Sublimation | and the contract | (c) | Respiration | (d) | Boiling | |
| 9. | A state function which describes together the internal energy and the product of pressure and | | | | | | | | | |
| | volume is called: | | | | | | | (GRW 11) | | |
| | (a) | Enthalpy | (b) | Internal energ | PY: | (c) | Wienk | (d) | Free energy | |

Multiple Choice Chemistry Questions On Thermochemistry

Gary S. Thorpe

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Therefore we need a forum where we can share and exchange information Seminars conferences and other scientific gatherings are the media through which we can do this Organizer Faculty of Mathematics and Natural Sciences of Universitas Negeri Medan Where Web Seminar via Zoom Meeting When Tuesday 8th November 2022 Theme The development of industrial based research in science and science education to improve research innovation strategy Topics AISTSSE 2020 included following topics 1 Mathematics Science 2 Mathematics Education 3 Physics Science 4 Physics Education 5 Biology Science 6 Biology Education 7 Chemistry Science 8 Chemistry Education 9 Computer Science 10 Science Education Scientific Committee 1 Prof Dr Syawal Gultom M Pd Universitas Negeri Medan Indonesia 2 Prof Dr Marleen Kamperman University of Groningen Netherland 3 Prof Manihar Situmorang M Sc Ph D Universitas Negeri Medan Indonesia 4 Prof Tsunenori Mine School of Engineering Department of Electrical Engineering and Computer Science Kyushu University Japan 5 Prof Dian Armanto M Pd Universitas Negeri Medan Indonesia 6 Prof Dr Herbert Sipahutar M Sc Universitas Negeri Medan Indonesia 7 Prof Abedel Karrem Nasser M Alomari Department of Mathematics Faculty of Science Yarmouk University Jordan 8 Prof Dr Bornok Sinaga M Pd Universitas Negeri Medan Indonesia 9 Prof Dr Muhammad Sattar Rasul Universitas Kebangsaan Malaysia Malaysia 10 Prof Motlan M Sc Ph D Universitas Negeri Medan Indonesia 11 Prof Dr Asmin M Pd Universitas Negeri Medan Indonesia 12 Prof Dr Fauziyah Harahap M Si Universitas Negeri Medan Indonesia 13 Prof Dr Mukhtar M Pd Universitas Negeri Medan Indonesia 14 Prof Dr Pargaulan Siagian M Pd Universitas Negeri Medan Indonesia 15 Prof Dr Sahat Saraqih M Pd Universitas Negeri Medan Indonesia 16 Prof Dr Edi Syahputra M Pd Universitas Negeri Medan Indonesia 17 Prof Dr Hasratuddin M Pd Universitas Negeri Medan Indonesia 18 Prof Dr Ramlan Silaban M Si Universitas Negeri Medan Indonesia 19 Prof Dr Retno Dwi Suyanti M Si Universitas Negeri Medan Indonesia 20 Prof Dr Nurdin Bukit M Si Universitas Negeri Medan Indonesia 21 Prof Dr Sahyar M S Universitas Negeri Medan Indonesia 22 Prof Dr rer nat Binari Manurung M Si Universitas Negeri Medan Indonesia 23 Prof Dr Makmur Sirait M Si Universitas Negeri Medan Indonesia 24 Prof Dr Eva Marlina Ginting M Si Universitas Negeri Medan Indonesia 25 Prof Dr Drs Tri Harsono M Si Universitas Negeri Medan Indonesia 26 Prof Dr Martina Restuati M Si Universitas Negeri Medan Indonesia 27 Prof Drs Zul Amry M Si Ph D Universitas Negeri Medan Indonesia Supported by FORUM MIPA LPTK INDONESIA **Kaplan PRAXIS.** Kaplan, 2007 Taking prospective teachers step by step through the entire preparation process Kaplan PRAXIS 2007 Edition offers 2 full length PPST practice tests 1 full length PLT test and 3 Elementary Educations tests In depth review for PRAXIS I and II including the Pre Professional Skills Test PPST Multiple Subjects Assessment for Teachers MSAT Principles of Learning and Teaching PLT and Elementary Education tests Effective score raising strategies Question banks for English Science Mathematics and Social Studies The ETS Test Collection Catalog Educational Testing Service. Test Collection, 1993 The major source of information on the availability of standardized tests Wilson Library Bulletin Covers commercially available standardized tests and hard to locate research instruments Learning Thermochemistry Karen

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