

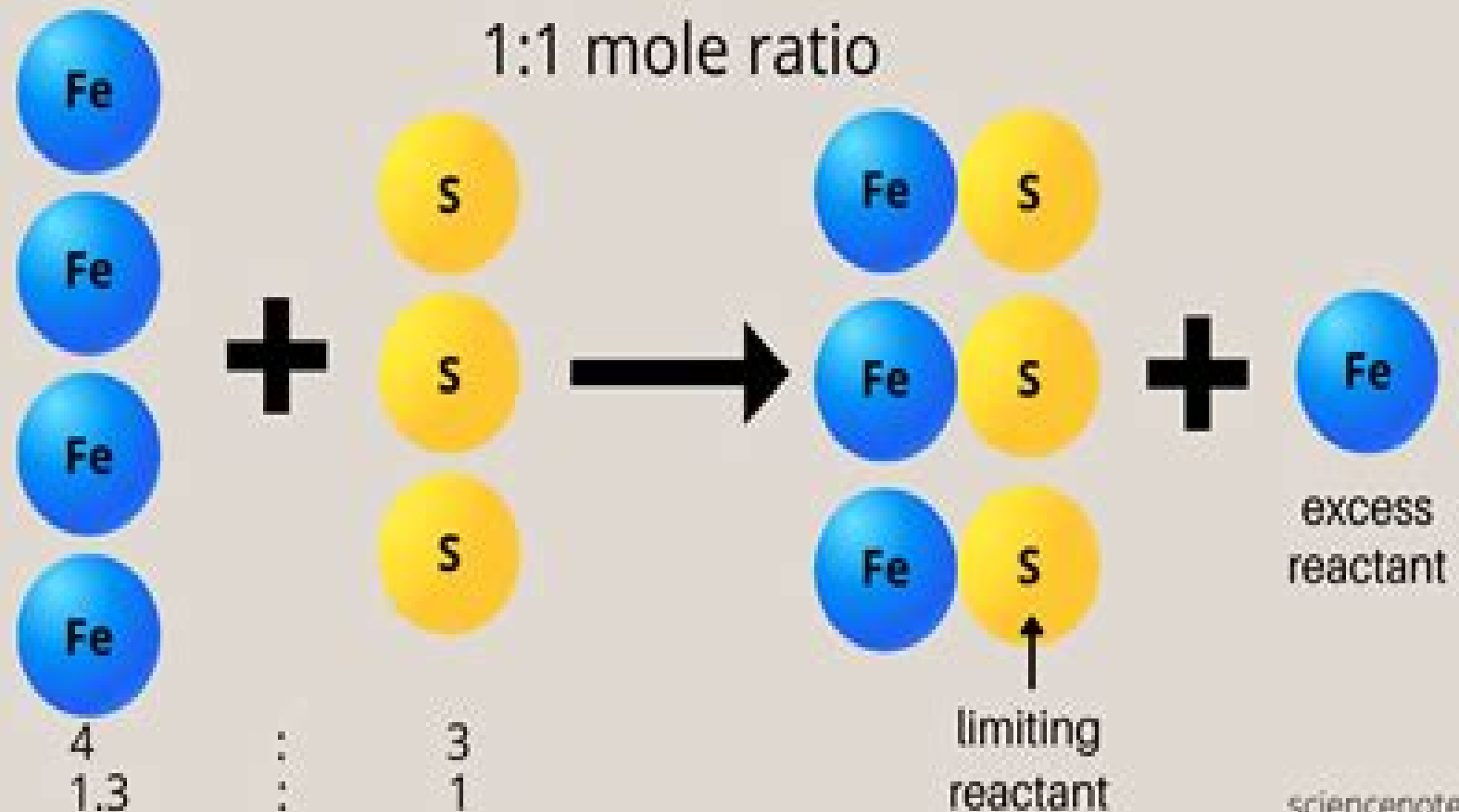
# Limiting Reactant

The limiting reactant is the reactant that is completely consumed in a chemical reaction.

Find it using the mole ratio in the balanced chemical equation.



1:1 mole ratio



# Limiting Reactants C1y

**Lifelique**



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*Limiting Reactant and Percent Lifeliqe*,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction     Handbook of Heat and Mass Transfer: Advances in reactor design and combustion science ,1990     **Derwent British Patents Report** ,1961

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