

Chemistry 105

Chapter 3 Review Problems

- How many molecules are present in 180 grams of water? (The atomic weights of H and O are 1 g/mol and 16 g/mol, respectively.)
a) 6.0×10^{24}
b) 10
c) 5.6×10^{-28}
d) 1.8×10^2
e) 180×10^{28}
- The atomic mass of sulfur is given as 32.07 g/mol. What is the mass of an average atom of sulfur?
a) 5.33×10^{-23} g
b) 32.07 g
c) 32.07×10^{-23} kg
d) 193×10^{-23} g
e) 6.02×10^{23} g
- What is the mass of C present in 60 g of C_3H_8 ?
a) 12 g
b) 24 g
c) 48 g
d) 60 g
e) 6 g
- What is the molar mass of C_3H_8O ?
a) 24 g/mol
b) 32 g/mol
c) 7 g/mol
d) 35 g/mol
e) 46 g/mol
- What is the correct percentage by mass of nitrogen in HCN?
a) 33 %
b) 14%
c) 27%
d) 52%
e) 98%
- What is the chief danger that is anticipated if CO_2 levels get too high in the atmosphere?
a) people will develop skin cancer from exposure to UV light
b) the atmosphere will develop an unpleasant odor
c) the temperature of the planet may climb by several degrees
d) plant life will be exterminated
e) The CO_2 will remove ozone from the stratosphere

Modern Chemistry Chapter 10 3 Review Answers

Zhe Xu



Modern Chemistry Chapter 10 3 Review Answers:

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In a world used by displays and the ceaseless chatter of instantaneous interaction, the melodic elegance and emotional symphony developed by the prepared term frequently fade in to the back ground, eclipsed by the relentless sound and disturbances that permeate our lives. But, nestled within the pages of **Modern Chemistry Chapter 10 3 Review Answers** a wonderful literary treasure brimming with raw emotions, lies an immersive symphony waiting to be embraced. Constructed by a wonderful composer of language, this charming masterpiece conducts viewers on a psychological journey, skillfully unraveling the hidden tunes and profound affect resonating within each carefully constructed phrase. Within the depths of the touching review, we will discover the book is key harmonies, analyze their enthralling publishing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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