

PRE TEST

Chapter 6 Test-General/Honors Chemistry

Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- | | | | |
|-------|--------------|---------------------------|--|
| C.2.2 | <u> a </u> | 1. cation | a. a positively charged ion |
| C.2.2 | <u> g </u> | 2. electronegativity | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled |
| C.2.5 | <u> j </u> | 3. inner transition metal | c. the energy required to remove an electron from an atom |
| C.2.6 | <u> i </u> | 4. alkali earth metal | d. good conductor of heat and electric current |
| C.2.6 | <u> h </u> | 5. noble gas | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons |
| C.2.5 | <u> b </u> | 6. representative element | f. any nonmetal in group 7A |
| C.2.2 | <u> d </u> | 7. metal | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u> f </u> | 8. halogen | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled |
| C.2.5 | <u> e </u> | 9. transition metal | i. an element in group 2A |
| C.2.2 | <u> c </u> | 10. ionization energy | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons |

Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- | | | | | |
|-------|--------------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u> b </u> | 11. The alkali metals do not include | a. Li | c. Na |
| | | | b. Ca | d. Rb |
| C.2.6 | <u> c </u> | 12. The element iodine, I, is a | a. period 5 alkali metal | c. period 5 halogen |
| | | | b. period 4 halogen | d. period 5 transition metal |

Modern Chemistry Chapter 1review Answers

WJ Hussar



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Table of Contents Modern Chemistry Chapter 1review Answers

1. Understanding the eBook Modern Chemistry Chapter 1review Answers
 - The Rise of Digital Reading Modern Chemistry Chapter 1review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 1review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 1review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 1review Answers
 - Personalized Recommendations
 - Modern Chemistry Chapter 1review Answers User Reviews and Ratings
 - Modern Chemistry Chapter 1review Answers and Bestseller Lists
5. Accessing Modern Chemistry Chapter 1review Answers Free and Paid eBooks
 - Modern Chemistry Chapter 1review Answers Public Domain eBooks
 - Modern Chemistry Chapter 1review Answers eBook Subscription Services
 - Modern Chemistry Chapter 1review Answers Budget-Friendly Options
6. Navigating Modern Chemistry Chapter 1review Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Modern Chemistry Chapter 1review Answers Compatibility with Devices
 - Modern Chemistry Chapter 1review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 1review Answers
 - Highlighting and Note-Taking Modern Chemistry Chapter 1review Answers
 - Interactive Elements Modern Chemistry Chapter 1review Answers
8. Staying Engaged with Modern Chemistry Chapter 1review Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter 1review Answers
9. Balancing eBooks and Physical Books Modern Chemistry Chapter 1review Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter 1review Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Chapter 1review Answers
- Setting Reading Goals Modern Chemistry Chapter 1review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Chapter 1review Answers
- Fact-Checking eBook Content of Modern Chemistry Chapter 1review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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