Chemistry

(www.tiwariacademy.com) (Chapter 13) (Amines) (Exercises) (Class 12)

Question 13.1:

Write IUPAC names of the following compounds and classify them into primary, secondary and tertiary amines.

(i) (CH₃)₂ CHNH₂ (iv) (CH₃)₃CNH₂ (ii) CH₃(CH₂)₂NH₂ (v) C₆H₅NHCH₃ (iii) CH₃NHCH(CH₃)₂ (vi) (CH₃CH₂)₂NCH₃

- (vii) m-BrC6H4NH2
- (i) 1-Methylethanamine (1° amine)
- (ii) Propan-1-amine (1° amine)
- (iii) N-Methyl-2-methylethanamine (2° amine)
- (iv) 2-Methylpropan-2-amine (1°amine)
- (v) N-Methylbenzamine or N-methylaniline (2° amine)
- (vi) N-Ethyl-N-methylethanamine (3° amine)
- (vii) 3-Bromobenzenamine or 3-bromoaniline (1° amine)

Question 13.2:

Give one chemical test to distinguish between the following pairs of compounds.

(i) Methylamine and dimethylamine

(ii) Secondary and tertiary amines

(iii) Ethylamine and aniline

(iv) Aniline and benzylamine

- (v) Aniline and N-methylaniline.
- Answer 13.2:
- (i) Methylamine and dimethylamine can be distinguished by the carbylamine test.

Carbylamine test: Aliphatic and aromatic primary amines on heating with chloroform andethanolic potassium hydroxide form foul-smelling isocyanides or carbylamines. Methylamine (being an aliphatic primary amine) gives a positive carbylamine test, but dimethylamine does not.

CH₃ − NH₂ + CHCl₃ + 3 KOH
$$\stackrel{\triangle}{\longrightarrow}$$
 CH₃− NC + 3KCl + 3H₂

Methylamine(1') Methylisocyanide (foul smell)

(CH₃)₂NH + CHCl₃ + 3KOH $\stackrel{\triangle}{\longrightarrow}$ No reaction

(ii) Secondary and tertiary amines can be distinguished by allowing them to react with Hinsberg's reagent (benzenesulphonyl chloride, C₆H₅SO₂Cl). Secondary amines react with Hinsberg's reagent to form a product that is insoluble in an alkali. For example, N, N-diethylamine reacts with Hinsberg's reagent to form N, N-diethylbenzenesulphonamide, which is insoluble in an alkali. Tertiary amines, however, do not react with Hinsberg's reagent.

Benzenesulphonyl chloride

N. N - Diethylbenzenesulphonamide

(iii) Ethylamine and aniline can be distinguished using the azo-dye test. A dye is obtained when aromatic amines react with HNO₂ (NaNO₂ + dil.HCl) at 0-5°C, followed by a reaction with the alkaline solution of 2-naphthol. The dye is usually yellow, red, or orange in colour. Aliphatic amines give a brisk effervescence due (to the evolution of N₂ gas) under similar conditions.

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