

## CHAPTER 2 WORKSHEET: CHEMISTRY OF LIFE

### 1. Chapter 2, Section 1: Nature of Matter

- a. Matter: anything that occupies space. Ex: Desk, Human, Air, ect.
- b. Atom: Smallest part of matter. It **can not** be broken down into smaller pieces. Composed of 3 particles: Protons (+), Electrons (-), and Neutrons (n)
- c. Element: Substance that is only made of one kind of atom. Ex: Hydrogen, Oxygen, ect.
  - i. C, H, N, O, P, S
- d. Compound: Two or more different atoms or elements joined together. Ex: Salt =
- e. Covalent Bond: Two or more atoms sharing electrons to form a molecule. Covalent bonds make molecules, ex: Water, Carbon Dioxide, and Oxygen Gas
- f. Hydrogen Bond: A weak chemical attraction between **polar** molecules. Ex: Two water molecules
- g. Ionic Bond: Gain or Loss of electrons that create a molecule
- h. Ion: An atom that has gained (-) electrons or (loss) electrons (+)

### 2. Chapter 2, Section 2: Water and Solutions

- a. How does water help maintain homeostasis?
  - i. Heats slowly
  - ii. Retains heat
  - iii. Carries Heat away through water vapor (sweating)
  - iv. Universal solvent (anything except for a polar substance can dissolve in it)
- b. What is the difference between adhesion & cohesion?
  - i. Adhesion: different substances
  - ii. Cohesion: alike substances
- c. What is the pH range for an Acid? Base?
  - i. Acid = 0 – 6.9
  - ii. Neutral = 7
  - iii. Base = 7.1 – 14
- d. What type of substance is an example of an acid? Of a base?
  - i. Acids: Lemon, Vinegar, Milk
  - ii. Bases: Household cleaning material

### 3. Chapter 2, Section 3: Chemistry of Cells

- a. Carbohydrates: Organic substances, made up of carbon, hydrogen and oxygen atoms. Always in the ratio of 1:2:1. Key source of energy.
  - i. They are made up of Monosaccharide: Single sugars (glucose)
  - ii. Disaccharides are double sugars: Sucrose
  - iii. Polysaccharides: Starch and cellulose

# Modern Chemistry Chapter 2 Mixed Review Worksheet

## Answer Key

**J Dewey**



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alcohols reactions The chapter Atomic Structure and Theory MCQs covers topics of atom facts elements and atoms number of nucleons protons electrons and neutrons The chapter Benzene Chemical Compound MCQs covers topics of benzene arenes reaction phenol properties and reactions of phenol The chapter Carbonyl Compounds MCQs covers topics of carbonyl compounds aldehydes and ketone testing nucleophilic addition with HCN preparation of aldehydes and ketone reduction of aldehydes and ketone

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