

balance nuclear reaction equations

aluminum-26 undergoes positron emission
potassium-40 undergoes beta decay (electron)
uranium-238 undergoes alpha decay



electron configurations

Write the electron configuration for C.

Write the electron configuration for Cr.

Write the shorthand notation for Bi.



NOMENCLATURE [naming compounds, writing formulas]

Write the formula for each of the following compounds:

- a. Sulfur hexafluoride SF_6
- b. Lithium nitride Li_3N
- c. Chromium (III) carbonate $\text{Cr}_2(\text{CO}_3)_3$
- d. Tin (II) chloride SnCl_2
- e. Ammonium acetate $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$ (or $\text{CH}_3\text{COONH}_4$)
- f. Mercury (I) chloride Hg_2Cl_2
- g. Potassium bromate KBrO_3
- h. Hydrosulfuric acid $\text{H}_2\text{S(aq)}$
- i. Chloric acid HClO_3
- j. Sulfurous acid H_2SO_3

Name each of the following compounds.

- a. CuSO_4 Copper (II) sulfate
- b. AlF_3 aluminum fluoride
- c. HI hydrogen iodide
- d. NO nitrogen monoxide
- e. $\text{H}_2\text{Se(aq)}$ hydroselenic acid
- f. HNO_3 nitric acid
- g. HNO_2 nitrous acid
- h. NaHSO_4 sodium hydrogen sulfate

Half-life

Tritium (H-3) is a radioactive isotope of hydrogen with a half-life of 12.3 years. How long would it take for a 40.0 g sample to decay down to 1.25 g?

$$40 \xrightarrow{1} 20 \xrightarrow{2} 10 \xrightarrow{3} 5 \xrightarrow{4} 2.5 \xrightarrow{5} 1.25$$

Fe-61 has a half-life of 6.00 min. Of a 100.0 mg sample, how much will remain after 18.0 min?

$$3 \text{ half lives} \quad 100 \xrightarrow{1} 50 \xrightarrow{2} 25 \xrightarrow{3} 12.5 \text{ mg}$$

Average atomic mass

The element copper has naturally occurring isotopes with mass numbers of 63 and 65.

The relative abundance and atomic masses are 69.2% for a mass of 63amu and 30.8% for a mass of 65amu. Calculate the average atomic mass of copper.

$$\begin{array}{rcl} 63 \times .692 & = & 43.596 \rightarrow 44 \\ 65 \times .308 & = & 20.02 \rightarrow 20 \\ + & & \\ \hline & & 63.62 \rightarrow 64 \end{array}$$

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**Holt Rinehart & Winston, Holt,
Rinehart and Winston Staff**



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Supplement for Modern Organic Chemistry John D. Roberts, Marjorie C. Caserio, 1967 **Principles of Modern Chemistry** David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom **Unlocking Nature: Foundations and Frontiers of Modern Science** Syed Feroze Ahamed, Dr. Raj Kumar Deshmukh, Kiran .K, A RENUKA , PREFACE The story of science is at its core the story of humanity s enduring curiosity From the first spark of wonder that lit in the minds of early hunter gatherers as they gazed at the night sky to the sophisticated instruments now probing the edges of black holes and sequencing the human genome the pursuit of knowledge has been a defining feature of our species This book Unlocking Nature Foundations and Frontiers of Modern Science is an exploration of that timeless quest its roots its revolutions and its future horizons For centuries humans have asked questions that at first seemed unanswerable What is the universe made of How did life begin Why do diseases emerge and can they be prevented How can we harness the forces of nature without destroying it Each era in history offered partial answers often wrapped in myth philosophy or limited observation Yet with every passing generation methods sharpened tools improved and explanations became more refined Science emerged not simply as a body of knowledge but as a method of thinking systematic evidence based and self correcting The aim of this book is not merely to recount scientific milestones but to present science as a living evolving dialogue between humanity and nature It seeks to inspire readers whether students professionals or curious minds to appreciate both the elegance of scientific principles and the responsibilities that come with them Science is not only about what we can do but also about what we ought to do I extend my gratitude to the countless scientists educators and visionaries whose work forms the backbone of this narrative and to the readers who carry forward the spirit of inquiry May this book serve as both a reflection on our journey so far and an invitation to imagine what lies ahead Unlocking Nature is ultimately about more than science It is about our shared human adventure the unfolding story of discovery challenge and wonder that defines our place in the cosmos Authors SYED FEROZE AHAMED DR RAJ KUMAR DESHMUKH KIRAN K A RENUKA Modern Physical Organic Chemistry Eric V.

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