Name Class

Assessment)

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- Which of the properties listed below is not characteristic of an acid?
 - a. a sour taste
 - b. a slippery feel
 - c. the ability to conduct an electric current
 - d. reactivity with metals
 - 2. Which is an example of a weak acid?
 - a. HCI
 - **b.** H₂SO₄
 - c. HaPO₄
 - d. HNO_a
- _____ 3. Acetic acid is found in
 - a. vinegar.
 - b. the stomach.
 - c. antacids.
 - d. oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
 - a. is an electrolyte.
 - b. tastes bitter.
 - c. dissociates to form OH ons.
 - d. All of the above
- 5. Which of the following is a Brønsted-Lowry base?
 - a. HCl
 - b. HCO_a
 - c. H₀O *
 - d. H_0PO_4

6. In the reaction represented by the equation

$$H_2C_2O_4(aq) + CH_3NH_2(aq) \Leftrightarrow HC_2O_4^-(aq) + CH_3NH_2^+(aq),$$

which of these is a conjugate acid-base pair?

- a. $H_2C_2O_4(aq)$ and $CH_3NH_2(aq)$
- b. H₂C₂O₄(aq) and CH₃NH^{*}₃(aq)
- c. CH₃NH₂(aq) and CH₃NH'₃(aq)
- d. HC2O4(aq) and CH2NH2(aq)

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Table of Contents Modern Chemistry Chapter 2 Test

- 1. Understanding the eBook Modern Chemistry Chapter 2 Test
 - The Rise of Digital Reading Modern Chemistry Chapter 2 Test
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Modern Chemistry Chapter 2 Test
 - Exploring Different Genres
 - o Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 2 Test
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Modern Chemistry Chapter 2 Test
 - Personalized Recommendations
 - Modern Chemistry Chapter 2 Test User Reviews and Ratings
 - Modern Chemistry Chapter 2 Test and Bestseller Lists

- 5. Accessing Modern Chemistry Chapter 2 Test Free and Paid eBooks
 - Modern Chemistry Chapter 2 Test Public Domain eBooks
 - Modern Chemistry Chapter 2 Test eBook Subscription Services
 - Modern Chemistry Chapter 2 Test Budget-Friendly Options
- 6. Navigating Modern Chemistry Chapter 2 Test eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Modern Chemistry Chapter 2 Test Compatibility with Devices
 - Modern Chemistry Chapter 2 Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 2 Test
 - Highlighting and Note-Taking Modern Chemistry Chapter 2 Test
 - o Interactive Elements Modern Chemistry Chapter 2 Test
- 8. Staying Engaged with Modern Chemistry Chapter 2 Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter 2 Test
- 9. Balancing eBooks and Physical Books Modern Chemistry Chapter 2 Test
 - Benefits of a Digital Library
 - \circ Creating a Diverse Reading Collection Modern Chemistry Chapter 2 Test
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Modern Chemistry Chapter 2 Test
 - Setting Reading Goals Modern Chemistry Chapter 2 Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Modern Chemistry Chapter 2 Test
 - Fact-Checking eBook Content of Modern Chemistry Chapter 2 Test
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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