

Chemical Bonding

Section 4: Ionic Bonding and Ionic Compounds

In this space, provide your answer to the title of this section of physics and then complete each statement or finish answers each question.

1. The most representative ion represented by transition metals is from
 - a. Group 1 (alkali metals)
 - b. Group 2 (alkaline earth metals)
 - c. Group 3 (transition metals)
 - d. Group 4 (transition metals)
2. Compared to the energy of ionization, the energy of hydration is
 - a. lower potential energy
 - b. higher potential energy
 - c. lower potential energy
 - d. equal potential energy
3. The most likely ion of transition metals is
 - a. the most representative ion
 - b. the most representative ion
 - c. the most representative ion
 - d. the most representative ion
4. Compared to the ionization energy of transition metals, the energy of hydration is
 - a. lower potential energy
 - b. higher potential energy
 - c. lower potential energy
 - d. equal potential energy
5. The most likely ion of transition metals is
 - a. the most representative ion
 - b. the most representative ion
 - c. the most representative ion
 - d. the most representative ion

Modern Chemistry Chapter 5 Section 1 Quiz Answers

IM Harris



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