

7-3 Review and Reinforcement

Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

| Number | Prefix |
|--------|--------|
| 1 | mono |
| 2 | di |
| 3 | tri |
| 4 | tetra |
| 5 | penta |
| 6 | hexa |
| 7 | hepta |
| 8 | octa |
| 9 | nona |
| 10 | deca |

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation
anion

hydrates
anhydrates

electronegative
hydrogen ions

ionic
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate**.
- When naming an ionic substance, the name of the **cation** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **electronegative** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **hydrogen ions**.
- The name of an acid is usually derived from the name of its characteristic **anion**.
- Prefixes are often used when naming **molecular** compounds.

Modern Chemistry Chapter 7 Section Review Answers

F Rizvi



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Table of Contents Modern Chemistry Chapter 7 Section Review Answers

1. Understanding the eBook Modern Chemistry Chapter 7 Section Review Answers
 - The Rise of Digital Reading Modern Chemistry Chapter 7 Section Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Modern Chemistry Chapter 7 Section Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Modern Chemistry Chapter 7 Section Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Modern Chemistry Chapter 7 Section Review Answers

- Personalized Recommendations
 - Modern Chemistry Chapter 7 Section Review Answers User Reviews and Ratings
 - Modern Chemistry Chapter 7 Section Review Answers and Bestseller Lists
5. Accessing Modern Chemistry Chapter 7 Section Review Answers Free and Paid eBooks
- Modern Chemistry Chapter 7 Section Review Answers Public Domain eBooks
 - Modern Chemistry Chapter 7 Section Review Answers eBook Subscription Services
 - Modern Chemistry Chapter 7 Section Review Answers Budget-Friendly Options
6. Navigating Modern Chemistry Chapter 7 Section Review Answers eBook Formats
- ePub, PDF, MOBI, and More
 - Modern Chemistry Chapter 7 Section Review Answers Compatibility with Devices
 - Modern Chemistry Chapter 7 Section Review Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Modern Chemistry Chapter 7 Section Review Answers
 - Highlighting and Note-Taking Modern Chemistry Chapter 7 Section Review Answers
 - Interactive Elements Modern Chemistry Chapter 7 Section Review Answers
8. Staying Engaged with Modern Chemistry Chapter 7 Section Review Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Modern Chemistry Chapter 7 Section Review Answers
9. Balancing eBooks and Physical Books Modern Chemistry Chapter 7 Section Review Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Modern Chemistry Chapter 7 Section Review Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Modern Chemistry Chapter 7 Section Review Answers
- Setting Reading Goals Modern Chemistry Chapter 7 Section Review Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Modern Chemistry Chapter 7 Section Review Answers

- Fact-Checking eBook Content of Modern Chemistry Chapter 7 Section Review Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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