



Chemical Reactions and Equations

Chemical reactions are the process in which new substance with new properties are formed.

The burning of magnesium in air to form magnesium oxide is an example of a chemical reaction



Note:- The magnesium ribbon which we use usually has a coating of 'magnesium oxide' on its surface which is formed by the slow action of oxygen of air on it. So before burning in air, the magnesium ribbon is cleaned by rubbing with a sand paper. This is done to remove the protective layer of magnesium oxide from the surface of magnesium ribbon so that it may readily combine with the oxygen of air.

Characteristics of Chemical Reactions

- (i) Evolution of a gas
- (ii) Formation of a precipitate
- (iii) Change in colour
- (iv) Change in temperature
- (v) Change in state

1.) Evolution of a gas

The chemical reaction b/w zinc and dilute sulphuric acid is characterized by the evolution of Hydrogen gas



Modern Chemistry Chemical Equations And Reactions Notes

WJ Hussar



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