

# Chemistry Worksheet BUNDLE



## Calculating Molar Mass

**Instructions:**  
Calculate the molar mass for each of the compounds below. Write your answers in the space provided.



Example:  $H_2O$

$H$  = 1 atomic mass unit (amu)  $O$  = 16 amu  $H_2O$  = 18 amu

Carbon dioxide	Formula: _____	Molar Mass: _____
Hydrogen peroxide	Formula: _____	Molar Mass: _____
Ammonia	Formula: _____	Molar Mass: _____
Hydrochloric acid	Formula: _____	Molar Mass: _____
Sulfuric acid	Formula: _____	Molar Mass: _____
Phosphoric acid	Formula: _____	Molar Mass: _____
Acetic acid	Formula: _____	Molar Mass: _____
Glucose	Formula: _____	Molar Mass: _____
Fructose	Formula: _____	Molar Mass: _____
Starch	Formula: _____	Molar Mass: _____
Cellulose	Formula: _____	Molar Mass: _____
Protein	Formula: _____	Molar Mass: _____
DNA	Formula: _____	Molar Mass: _____
RNA	Formula: _____	Molar Mass: _____
Lipids	Formula: _____	Molar Mass: _____
Carbohydrates	Formula: _____	Molar Mass: _____
Enzymes	Formula: _____	Molar Mass: _____
Monomers	Formula: _____	Molar Mass: _____
Polymers	Formula: _____	Molar Mass: _____
Alloys	Formula: _____	Molar Mass: _____
Composites	Formula: _____	Molar Mass: _____
Metals	Formula: _____	Molar Mass: _____
Nonmetals	Formula: _____	Molar Mass: _____
Metalloids	Formula: _____	Molar Mass: _____
Isotopes	Formula: _____	Molar Mass: _____
Radioactive elements	Formula: _____	Molar Mass: _____
Stable elements	Formula: _____	Molar Mass: _____
Unstable elements	Formula: _____	Molar Mass: _____
Artificial elements	Formula: _____	Molar Mass: _____
Natural elements	Formula: _____	Molar Mass: _____
Trace elements	Formula: _____	Molar Mass: _____
Essential elements	Formula: _____	Molar Mass: _____
Nonessential elements	Formula: _____	Molar Mass: _____
Toxic elements	Formula: _____	Molar Mass: _____
Beneficial elements	Formula: _____	Molar Mass: _____
Harmful elements	Formula: _____	Molar Mass: _____
Useful elements	Formula: _____	Molar Mass: _____
Dangerous elements	Formula: _____	Molar Mass: _____
Safe elements	Formula: _____	Molar Mass: _____
Radioactive isotopes	Formula: _____	Molar Mass: _____
Stable isotopes	Formula: _____	Molar Mass: _____
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Useful isotopes	Formula: _____	Molar Mass: _____
Dangerous isotopes	Formula: _____	Molar Mass: _____
Safe isotopes	Formula: _____	Molar Mass: _____

**ANSWER  
SHEETS!**



## Percent Composition

**Instructions:**  
Calculate the percent composition of each element in the compounds below. Write your answers in the space provided.



Example:  $H_2O$

$H$  = 1 amu  $O$  = 16 amu  $H_2O$  = 18 amu

Carbon dioxide	Formula: _____	% C: _____
Hydrogen peroxide	Formula: _____	% H: _____
Ammonia	Formula: _____	% N: _____
Hydrochloric acid	Formula: _____	% H: _____
Sulfuric acid	Formula: _____	% S: _____
Phosphoric acid	Formula: _____	% P: _____
Acetic acid	Formula: _____	% C: _____
Glucose	Formula: _____	% C: _____
Fructose	Formula: _____	% C: _____
Starch	Formula: _____	% C: _____
Cellulose	Formula: _____	% C: _____
Protein	Formula: _____	% C: _____
DNA	Formula: _____	% C: _____
RNA	Formula: _____	% C: _____
Lipids	Formula: _____	% C: _____
Carbohydrates	Formula: _____	% C: _____
Enzymes	Formula: _____	% C: _____
Monomers	Formula: _____	% C: _____
Polymers	Formula: _____	% C: _____
Alloys	Formula: _____	% C: _____
Composites	Formula: _____	% C: _____
Metals	Formula: _____	% C: _____
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Radioactive elements	Formula: _____	% C: _____
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Unstable isotopes	Formula: _____	% C: _____
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Natural isotopes	Formula: _____	% C: _____
Trace isotopes	Formula: _____	% C: _____
Essential isotopes	Formula: _____	% C: _____
Nonessential isotopes	Formula: _____	% C: _____
Toxic isotopes	Formula: _____	% C: _____
Beneficial isotopes	Formula: _____	% C: _____
Harmful isotopes	Formula: _____	% C: _____
Useful isotopes	Formula: _____	% C: _____
Dangerous isotopes	Formula: _____	% C: _____
Safe isotopes	Formula: _____	% C: _____

**25  
PROBLEMS**

# Calculating Molar Mass and Percent Composition



# Molar Mass And Percent Composition Answer Key

**S Marginson**



**Molar Mass And Percent Composition Answer Key:**

## **Molar Mass And Percent Composition Answer Key** Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Molar Mass And Percent Composition Answer Key**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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## **Molar Mass And Percent Composition Answer Key Introduction**

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