

Molarity Lab – Murder Investigation

BACKGROUND: A crime has been committed and now, as the forensic detective, we going to use your knowledge of solution chemistry to solve the crime. Miss Scarlet was found dead on the floor of the bathroom. Besides her body police found a clear liquid, undoubtedly the murder weapon. Detectives were quick to ascertain that the murder weapon was one of two chemicals. They knew that Miss Scarlet has a violent allergic reaction to potassium iodide, and that there was widespread use of silver nitrate, which is toxic if ingested, by several members of the household. Police further discovered the following evidence.

Professor Plum, an academic chemistry teacher, was working in the study with silver nitrate solutions of molarities 0.20 to 0.50. Miss Scarlet was Plum's worst student, arriving in class late, being responsible for lots of broken glassware, and rarely cleaning up after her experiments. The policeman in charge, never having had a chemistry class, did not seriously consider this to be a motive for murder.

Mrs. White, an astronomer, has a prescription for potassium iodide in the bathroom. Her pharmacist assures us that her potassium iodide solution has a molarity in the range of 0.05 M to 0.15 M. That's fine. Why was Professor Pluming with Miss Scarlet's on-boyfriend led the police to suspect her. Neither she nor her boyfriend could be found in the house for comment.

Mr. Green, a photographer, had solutions of silver nitrate between 0.05 M and 0.10 M in the laboratory. Mr. Green apparently had been caught red-handed at his own blackmail game. Pictures of him with Miss Scarlet were found hidden in Scarlet's bathroom. Mrs. Green should be spared seeing the pictures if at all possible.

Mrs. Pearsall, wanting the same class as Miss Scarlet, was found in the study, administering potassium iodide solution to her horse. When questioned, she admitted that her horse had a severe case of hemorrhoids. Her veterinarian informed detectives that the strength of the solution was greater than 0.20 M. It is likely that she knew of Miss Scarlet's violent allergy to potassium iodide, since they had been his partners in Professor Plum's chemistry class. Professor Plum recalled how angry Mrs. Pearsall would get when Miss Scarlet came late to class, irresponsible for the experiments, and how she inevitably got sick a few days before the lab was due.

Colonel Mustard had silver nitrate solutions in storage of 0.50 M, in the conservatory. The Colonel, who displayed the left-wing politics of Miss Scarlet, enjoyed his leisure hours making explosives. Colonel Mustard had been the senior assistant principal at the high school and was largely responsible for closing the campus when Miss Scarlet, then president of the student body, overthrew the administration and took control of the school.

YOUR TASK: You can obtain a sample of the murder weapon found at Miss Scarlet's side by asking your teacher for it, in the classroom. If you can determine whether it is silver nitrate or potassium iodide, then you can narrow the list of suspects considerably. And if you can determine the molarity of the unknown solution, then you can solve the case.

What was the murder? What was the murder weapon? What was its molarity? Provide a short, formal lab report giving evidence and analysis useful for the prosecution of the murderer. Your evidence will be Exhibit A at the trial, and will be subjected to the scrutiny of the defense attorney. On the basis of your evidence, a murderer might be set free or an innocent person executed.

Molarity Activity Murder Investigation Answers

Nicholas A. Bowman



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