MOLARITY BY DILUTION

79+52 = 162 Name _____

Acids are usually acquired from chemical supply houses in concentrated form. These acids are diluted to the desired concentration by adding water. Since moles of acid before dilution = moles of acid after dilution, and moles of acid = $M \times V$ then, $M_1 \times V_2 = M_2 \times V_3$. Solve the following problems.

 How much concentrated 18 M sulfuric acid is needed to prepare 250 mL of a 6.0 M solution?

83 mL]

 How much concentrated 12 M hydrochloric acid is needed to prepare 100,mL of a 2.0 M solution?

- 17mL/
- To what volume should 25 mL of 15 M nitric acid be diluted to prepare a 3.0 M solution?

4. To how much water should 50, mL of 12 M hydrochloric acid be added to produce a 4.0 M solution?

100ml H20)

5. To how much water should 100. mL of 18 M sulfuric acid be added to prepare a 1.5 M solution?

Molarity Instructional Fair Key

Tao Wei

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In a global used by displays and the ceaseless chatter of fast conversation, the melodic elegance and psychological symphony produced by the written word usually diminish into the back ground, eclipsed by the persistent sound and disturbances that permeate our lives. However, located within the pages of **Molarity Instructional Fair Key** a charming literary treasure overflowing with natural emotions, lies an immersive symphony waiting to be embraced. Crafted by a masterful composer of language, this captivating masterpiece conducts viewers on an emotional trip, well unraveling the hidden tunes and profound influence resonating within each cautiously constructed phrase. Within the depths of the touching review, we can investigate the book is main harmonies, analyze their enthralling publishing model, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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