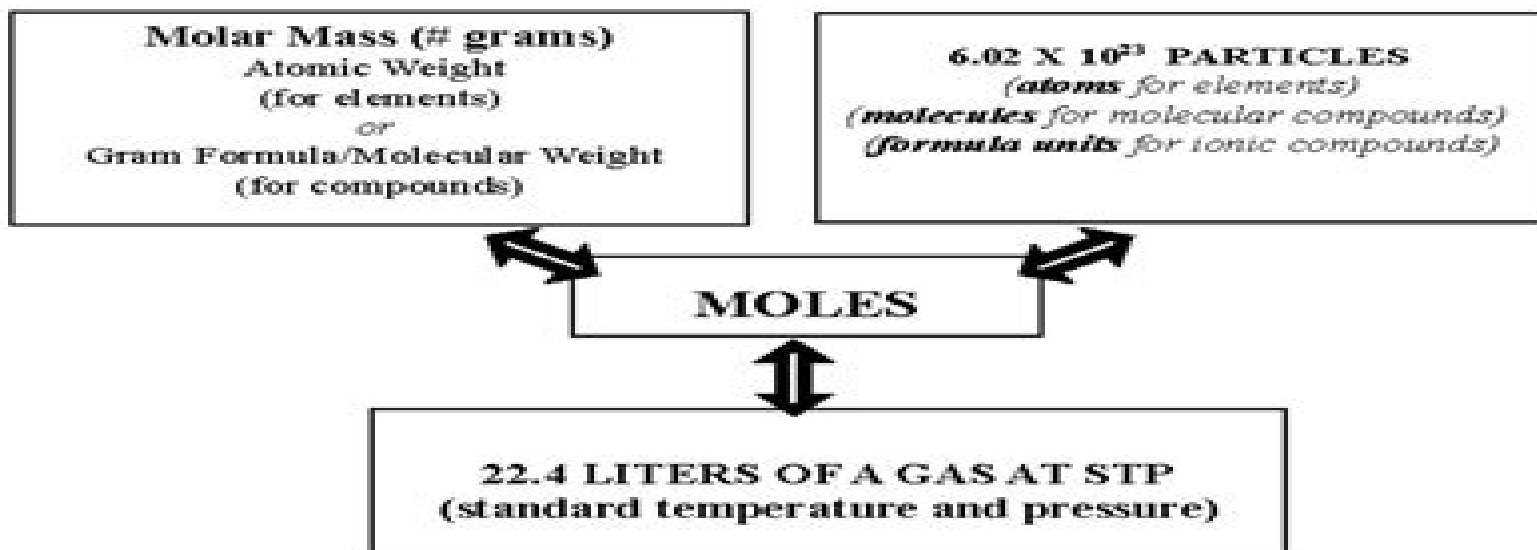


We are about to start on a unit of chemical calculations called "stoichiometry". Stoichiometry is how we calculate the relationships between the amounts of reactants and the amounts of products. For example, if we know the amount of reactants we have, we can use stoichiometry to calculate how many products the chemical reaction will produce.

This packet provides an organized, step-by-step approach for these problems. If you follow this system and complete work each day as it is assigned, this will not be a difficult unit. That is good, because stoichiometry is one of the most central and important concepts in chemistry. It is essential that you understand this unit in order to move forward in chemistry. In short: do not fall behind or you will be lost.

These problems involve numbers but no difficult mathematics. All you will ever have to do is add, multiply or divide. You will be expected to have a functioning calculator with you for every chemistry class.

The only new concept we will introduce in this unit is the idea of a mole. A mole is a quantity of matter that we use for conversion purposes. We can convert from grams to moles, liters to moles (for gases), and atoms, formula units, or molecules to moles. If you can convert any of these things to moles (and therefore moles to any of these things) we can convert grams to liters or molecules, liters to grams of molecules, and molecules to liters or grams.



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