

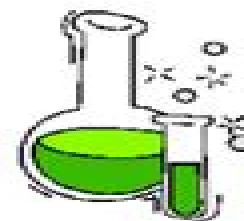
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The Mole Lab

Chemistry I Acc

(Weighing as a Means of Counting)



Introduction

One of the seven SI base units is the mole. The mole, also known as Avogadro's number, is equal to 6.02×10^{23} . The mole is a quantity like a dozen (12) or a gross (144). If you wanted to know how many eggs were in 3 dozen eggs you would multiply 3 dozen eggs \times 12 eggs/dozen. If you wanted to know how many firecrackers were in 3 gross firecrackers you would multiply 3 gross firecrackers \times 144 firecrackers/gross. Likewise if you wanted to know how many atoms were in 3 moles of atoms you would multiply 3 moles of atoms \times 6.02×10^{23} atoms/mole.

The Atomic Mass Unit is defined as 1/12 of a Carbon-12 atom. It is the standard in the AMU scale. The mass of all atoms in AMUs is defined relative to this standard. Likewise, the mass of a mole is defined as the number in grams that is equal to the average AMU of the element. For example the average atomic mass of Carbon is 12.011 AMU. The mass of one mole of Carbon is equal to 12.011 grams. The mass of one mole of Lead is equal to 207.2 grams. The mass of one mole of a molecule is equal to the mass of one mole of each of the atoms in the compound. For example the mass of one mole of Silicon dioxide is equal to 28.0855 grams for the Silicon plus 15.9994 grams for each of the oxygens (To make our lives easy we will round masses of all elements to the nearest 0.1 grams). Therefore, mass of one mole of SiO_2 is equal to 60.1 grams ($28.1 + 16.0 + 16.0$ grams). The mass of one mole of substance (element or compound) is called the substance's molar mass in units of grams/mole.

In part one of this lab you will convert between mass, moles and atoms of a given element and identify a substance based on calculating its molar mass. In part two you will use measurements of mass in order to convert between mass, moles, and molecules or atoms.

Materials

Part One:

Nine vials of various elements
Triple Beam Balance

Part Two:

1 canister Sodium Chloride
1 canister Calcium Carbonate
1 2 oz. Cup for Water
1 weighing boat for Sodium Chloride
1 weighing boat for Calcium Carbonate
1 stirrer
2 spoons

Mole Lab Chemistry I Acc

Richard Bailey



Mole Lab Chemistry I Acc:

Chemistry in the Laboratory James M. Postma, Julian L. Robert, J. Leland Hollenberg, 2004-03-12 This clearly written class tested manual has long given students hands on experience covering all the essential topics in general chemistry Stand alone experiments provide all the background introduction necessary to work with any general chemistry text This revised edition offers new experiments and expanded information on applications to real world situations Cooperative Chemistry Lab Manual Cooper, 2005-02 The laboratory course described in the lab manual emphasizes experimental design data analysis and problem solving Inherent in the design is the emphasis on communication skills both written and oral Students work in groups on open ended projects in which they are given an initial scenario and then asked to investigate a problem There are no formalized instructions and students must plan and carry out their own investigations Study Guide for Hein and Arena's Foundations of College Chemistry Peter Scott, Rachael Enriquez Porter, 1999-04-18 The Journal of Industrial and Engineering Chemistry ,1922 **Journal of Industrial and Engineering Chemistry** ,1922 *Books for Children* ,1960 Nuclear Science Abstracts ,1965 **Science** John Michels (Journalist), 1907 Vols for 1911 13 contain the Proceedings of the Helminothological Society of Washington ISSN 0018 0120 1st 15th meeting **Nuclear Science Abstracts** ,1972-06 **Industrial & Engineering Chemistry** ,1923 **Physics Briefs** ,1993 **Energy Research Abstracts** ,1987 **Journal of the Chemical Society** ,1905 Journal - Chemical Society, London Chemical Society (Great Britain), 1905 **Journal of the Chemical Society** Chemical Society (Great Britain), 1905 Titles of chemical papers in British and foreign journals included in Quarterly journal v 1 12 U.S. Government Research & Development Reports ,1967 **Journal of the Indian Chemical Society** Indian Chemical Society, 1995 **Bibliography of Scientific and Industrial Reports** ,1965-07 **American Druggist and Pharmaceutical Record** ,1898 *The Chemical News and Journal of Industrial Science* ,1913

Decoding **Mole Lab Chemistry I Acc**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Mole Lab Chemistry I Acc**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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