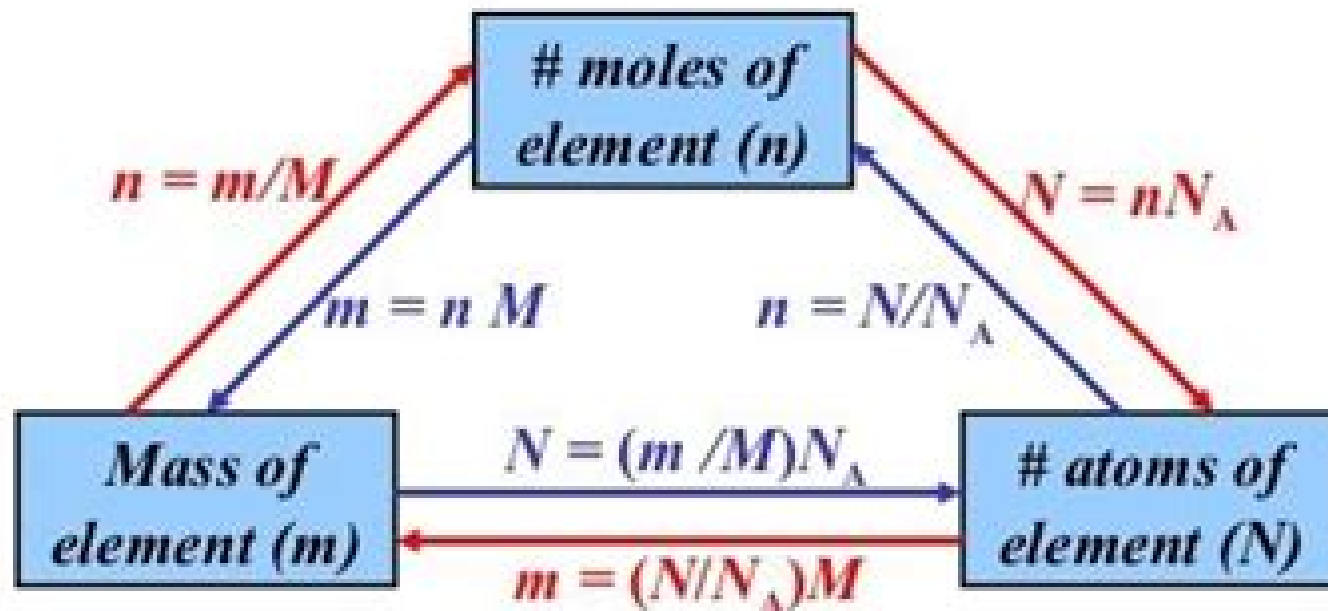


# Relationships between Mass, Mole, and Avogadro's Number



$n$ : number of moles

$m$ : mass

$M$ : molar mass

$N$ : number of atoms

$N_A$ : Avogadro's number,  $6.022 \times 10^{23}$

# Mole And Avogadros Number Answer Key

**NARAYAN CHANGDER**



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