

# Chapter 11

## Molecular Composition of Gases

### 11-1 Volume-Mass Relationships of Gases

Avogadro's Law, STP, molar volume

### 11-2 The Ideal Gas Law

$PV=nRT$  ( $R = 0.0821 \text{ (L}\cdot\text{atm)/(mol}\cdot\text{K)}$ ), finding molar mass and density

### 11-3 Stoichiometry of Gases

### 11-4 Effusion and Diffusion

Graham's law  $(\sqrt{M_B})/(\sqrt{M_A})$  (assuming constant T and P)

# Molecular Composition Of Gases 11 2 Answers

**Kathleen Armour**



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