Chapter 11 Molecular Composition of Gases

11-1 Volume-Mass Relationships of Gases Avogadro's Law, STP, molar volume

11-2 The Ideal Gas Law PV=nRT (R = 0.0821 (L·atm)/(mol·K)), finding molar mass and density

11-3 Stoichiometry of Gases

11-4 Effusion and Diffusion Graham's law $(\sqrt{M_B})/(\sqrt{M_A})$ (assuming constant T and P)

Molecular Composition Of Gases 11 3 Answers

Hussin A.Rothana

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