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# MOLES, GRAMS AND PARTICLES

## QUESTIONS AND ANSWERS

What is the mass, in grams, of 6.11 mol of sulfur trioxide?

- A) 80.1 g
- B) 48.6 g
- C) 391 g
- D) 489 g
- E) 801 g - answer--489 g

One mol of particles of any substance contains how many particles?

- A)  $10^6$
- B)  $3 \times 10^{-18}$
- C)  $3 \times 10^{10}$
- D)  $6.022 \times 10^{23}$
- E)  $6.022 \times 10^{-23}$  - answer-- $6.022 \times 10^{23}$

Avogadro's number is the number of

- A) particles in 1 mol of a substance
- B) amu in 1 mol of a substance
- C) grams in 1 mol of a substance
- D) moles in  $6.022 \times 10^{23}$  grams of an element
- E) moles in  $6.022 \times 10^{23}$  amu of an element - answer--particles in 1 mol of a substance

What is the mass of 0.00100 mol of caffeine,  $C_8H_{10}N_4O_2$ ?

- A) 194 g
- B) 87.0 g
- C) 0.194 g
- D) 0.870 g
- E) 1.94 g - answer--0.194 g

What is the mass of 2.00 mol of bromine, a liquid element?

- A) 16.0 g
- B) 79.9 g
- C) 70.0 g
- D) 160. g
- E) 320. g - answer--320. g

0.100 mol of lithium has a mass of

# Moles And Particles Answers

**Effiong Eyo**

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that challenges can be dealt with most efficient and better relationships can be developed among the youth

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authors tichaona mapolisa zimbabwe open university abstract this study sought to

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