

Chemistry -- Unit 6 Worksheet 1

After having observed molecular models which show 18.098 compounds, you discovered that certain particles are known to form positively charged ions, but cannot, within their natural particles, form negatively charged ions. (Lithium, Sodium, Potassium, Calcium, Barium, Strontium, Magnesium, Beryllium, and Radium). After having observed the 18.098 compounds, you also observed that certain particles are known to form negatively charged ions, but cannot, within their natural particles, form positively charged ions. (Fluorine, Chlorine, Bromine, Iodine, and Astatine). After having observed the particles that form the 18.098 compounds, you also observed that certain particles are known to form positively charged ions, but cannot, within their natural particles, form negatively charged ions. (Lithium, Sodium, Potassium, Calcium, Barium, Strontium, Magnesium, Beryllium, and Radium).

1. After having observed the particles that form the 18.098 compounds, you also observed that certain particles are known to form positively charged ions, but cannot, within their natural particles, form negatively charged ions. (Lithium, Sodium, Potassium, Calcium, Barium, Strontium, Magnesium, Beryllium, and Radium).

| 18.098 molecular formula | 1. Lithium 1. Sodium | 2. Potassium 2. Calcium | 3. Barium 3. Strontium | 4. Magnesium 4. Beryllium |
|--------------------------------|-------------------------|----------------------------|---------------------------|------------------------------|
| | $LiCl$ | $NaCl$ | KCl | $BeCl_2$ |

particle
diagram



| 18.098 molecular formula | 5. Barium 5. Strontium | 6. Magnesium 6. Calcium | 7. Sodium 7. Potassium |
|--------------------------------|---------------------------|----------------------------|---------------------------|
| | $BaCl_2$ | $MgCl_2$ | $NaCl$ |

particle
diagram



| 18.098 molecular formula | 8. Strontium 8. Barium | 9. Potassium 9. Sodium | 10. Magnesium 10. Calcium | 11. Barium 11. Strontium |
|--------------------------------|---------------------------|---------------------------|------------------------------|-----------------------------|
| | $SrCl_2$ | KCl | $MgCl_2$ | $BaCl_2$ |

particle
diagram



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Victor M. Corman



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