

Chapter Test B

Chapter: Chemical Equations and Reactions

Write a correct name (including oxidation state) of the react or product described. Completed tests must include correct answers to all questions.

- _____ 1. The product of a slightly soluble substance reacting in a double displacement reaction is often the formation of a
 - a. gas
 - b. precipitate**
 - c. molecular solution
 - d. solution
- _____ 2. An ionic compound equation is represented by which the
 - a. products**
 - b. reactants
 - c. formation of the products
 - d. transfer of products
- _____ 3. An ionic equation shows the activity of aq ions in the reaction
 - a. water
 - b. dissolved in water**
 - c. solid in aq
 - d. molecular
- _____ 4. The following list of replacement reactions are correct. Which is not?
 - a. Silver metal reacts with chlorine gas to produce silver chloride
 - b. Hydrogen gas reacts with oxygen gas to produce water
 - c. Sodium metal reacts with water to produce sodium hydroxide
 - d. Copper metal reacts with silver nitrate to produce silver metal and copper nitrate**
- _____ 5. The coefficients are balanced equations
 - a. indicate the number of moles of each substance that react**
 - b. show the number of grams of each substance that react
 - c. indicate the number of moles of the substances
 - d. show the relative volumes of each atom
- _____ 6. If species X is known that metal Y is less active than X, then
 - a. X will replace ions of Y in solution
 - b. Y will replace ions of X in solution**
 - c. Y will replace ions of X in solution
 - d. X will replace ions of Y in solution

Modern Chemistry Chapter Test B Answers

**David W. Oxtoby, H. Pat Gillis, Laurie J.
Butler**



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