

**CHAPTER 22 REVIEW*****Organic Chemistry*****MIXED REVIEW****SHORT ANSWER** Answer the following questions in the space provided.

1. \_\_\_\_\_ A saturated organic compound
- (a) contains all single bonds.
  - (b) contains at least one double or triple bond.
  - (c) contains only carbon and hydrogen atoms.
  - (d) is quite soluble in water.
2. Arrange the following in order of increasing boiling point:
- \_\_\_\_\_ a. ethane
  - \_\_\_\_\_ b. pentane
  - \_\_\_\_\_ c. heptadecane
3. Recall that isomers in organic chemistry have identical molecular formulas but different structures and IUPAC names.
- \_\_\_\_\_ a. Two isomers must have the same molar mass. True or False?
  - \_\_\_\_\_ b. Two isomers must have the same boiling point. True or False?
4. Explain why hydrocarbons with only single bonds cannot form geometric isomers.

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5. Write the IUPAC name for the following structural formulas:



# Modern Chemistry Ch 14 Ions Review Answers

**JA Banks**



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