

7-3 Review and Reinforcement

Naming Chemical Compounds

Write the correct numerical prefix for each of the numbers in the table below.

Number	Prefix
1	mono
2	di
3	tri
4	tetra
5	penta
6	hexa
7	hepta
8	octa
9	nona
10	deca

Complete each of the following sentences by filling in the appropriate word or phrase from the list below.

cation
anion

hydrates
anhydrates

electronegative
hydrogen ions

ionic
molecular

- Ionic compounds that absorb water into their solid structures form substances called **—hydrate**.
- When naming an ionic substance, the name of the **cation** comes first.
- To name a binary molecular compound, add the suffix *-ide* to the more **electronegative** of the two elements.
- By one definition, an acid is a molecular compound that dissolves in water to produce **hydrogen ions**.
- The name of an acid is usually derived from the name of its characteristic **anion**.
- Prefixes are often used when naming **molecular** compounds.

Modern Chemistry Chapter 7 Section 3 Review Answers

J Ma



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Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 *Study Guide to accompany Basic Concepts of Chemistry, 7th Edition* Leo J. Malone, 2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone's Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry **Independent and Weekly Review**, 1914 Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000

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