

Chemical Basis of Life

Module A Anchor 2

Key Concepts:

- Water is a polar molecule. Therefore, it is able to form multiple hydrogen bonds, which account for many of its special properties.
- Water's polarity gives it the ability to dissolve both ionic compounds and other polar molecules.
- Carbon can bond with many elements, including hydrogen, oxygen, phosphorus, sulfur, and nitrogen to form the molecules of life.
- The function of macromolecules is directly related to their chemical structure.
- Living things use carbohydrates as their main source of energy. Plants, some animals, and other organisms also use carbohydrates for structural purposes.
- Lipids can be used to store energy. Some lipids are important parts of biological membranes and waterproof coverings.
- Nucleic acids store and transmit hereditary, or genetic, information.
- Some proteins control the rate of reactions and regulate cell processes. Some proteins build tissues such as bone and muscle. Others transport materials or help fight disease.
- Chemical reactions always involve changes in the chemical bonds that join atoms in compounds.
- Chemical reactions that release energy often occur spontaneously. Chemical reactions that absorb energy will not occur without a source of energy.
- Enzymes speed up chemical reactions that take place in cells. This function is directly related to their structure, with each enzyme being specifically shaped to catalyze one particular reaction. Loss of structure causes loss of function.
- Temperature, pH, and regulatory molecules can affect the activity of enzymes.

Vocabulary:

Hydrogen bond

Adhesion

Heat of vaporization

Carbohydrate

Triglyceride

Fatty acid

Secondary structure

Enzyme

Substrate

solution

polarity

polymer

monosaccharide

phospholipid

protein

tertiary structure

catalyst

product

pH scale

freezing point

nucleic acid

polysaccharide

steroid

amino acid

quaternary structure

activation energy

reactant

cohesion

monomer

nucleotide

lipid

wax

primary structure

chemical reaction

active site

Properties of Water:

1. Describe the following properties of water and explain how each is important to living things: cohesion, adhesion, polarity, heat of vaporization, freezing point.

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S Ben Porath



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