

Molarity Worksheet #1

name _____

1. What does molarity mean?

$$\frac{\text{Number of moles of solute}}{1 \text{ liter solution}}$$

2. What is the molarity of a solution that contains 4.53 moles of lithium nitrate in 2.85 liters of solution?

$$\frac{4.53 \text{ mol LiNO}_3}{2.85 \text{ L soln}} = 1.59 \text{ M LiNO}_3$$

3. What is the molarity of a solution that contains 0.00372 moles hydrochloric acid in 2.39×10^{-2} liters of solution?

$$\frac{0.00372 \text{ mol HCL}}{2.39 \times 10^{-2} \text{ L soln}} = 0.156 \text{ M HCL}$$

4. A flask contains 85.5 g $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ (sucrose) in 1.00 liters of solution. What is the molarity?

$$\frac{85.5 \text{ g sucrose} \times 1 \text{ mol sucrose}}{1.00 \text{ L soln} \quad 342.34 \text{ g sucrose}} = 0.250 \text{ M sucrose}$$

5. A beaker contains 214.2 grams osmium (III) fluoride in 0.0673 liters of solution. What is the molarity?

$$\frac{214.2 \text{ g OsF}_3 \times 1 \text{ mol OsF}_3}{0.0673 \text{ L soln} \quad 247.23 \text{ g OsF}_3} = 12.9 \text{ M OsF}_3$$

6. Calculate the molarity if a flask contains 1.54 moles potassium sulfate in 125 ml of solution.

$$\frac{1.54 \text{ mol K}_2\text{SO}_4}{0.125 \text{ L soln}} = 12.3 \text{ M K}_2\text{SO}_4$$

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The Practice of Chemistry Donald J. Wink, Sharon Fetzner-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. The Practice of Chemistry First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools, and integrated media makes The Practice of Chemistry the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink. http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB

Linne & Ringsrud's Clinical Laboratory Science - E-Book Mary Louise Turgeon, 2015-02-10 Using a discipline by discipline approach, Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework. NEW Instrumentation updates show new technology being used in the lab. NEW Additional key terms in each chapter cover need-to-know terminology. NEW Additional tables and figures in each chapter clarify clinical lab science concepts.

Merrill Chemistry Robert C. Smoot, Richard G. Smith, Jack Price, 1998

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Clinical Laboratory Science - E-Book Mary Louise Turgeon, 2022-09-14 Selected for Doody's Core Titles 2024 in Laboratory Technology. Using a discipline by discipline approach, Turgeon's Clinical Laboratory Science: Concepts, Procedures, and Clinical Applications, 9th Edition provides a fundamental overview of the concepts, procedures, and clinical applications essential for working in a clinical laboratory and performing routine clinical lab tests. Coverage includes basic laboratory techniques and key topics such as safety, phlebotomy, quality assessment, automation, and point-of-care testing, as well as discussion of clinical laboratory specialties. Clear, straightforward instructions simplify laboratory procedures and are guided by the latest practices and CLSI Clinical and Laboratory Standards Institute standards. Written by well-known CLS educator Mary Louise Turgeon, this edition offers essential guidance and recommendations for today's laboratory testing methods and clinical applications. Broad scope of coverage makes this text an ideal companion for clinical

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chemistry for students aiming for careers in professional health care fields The Software Encyclopedia ,1988

Chemistry and Chemical Reactivity Mary L. Kotz,1996 **Proceedings of Annual Meeting [of The] National Association of Boards of Pharmacy and American Association of Colleges of Pharmacy, District One** National Association of Boards of Pharmacy. District 1,1963 **Resources in Education** ,1980

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