## Mole Ratios

How can the coefficients in a chemical equation be interpreted?

## Why?

A balanced chemical equation can tell us the number of reactant and product particles (ions, atoms, molecules or formula units) that are necessary to conserve mass during a chemical reaction. Typically when we balance the chemical equation we think in terms of individual particles. However, in real life the reaction represented by an equation occurs an unimaginable number of times. Short of writing very large numbers (10% or larger) in front of each chemical in the equation, how can we interpret chemical equations so that they more realistically represent what is happening in real life? In this activity you will explore the different ways a chemical reaction can be interpreted.

#### Model 1 – A Chemical Reaction

$$N_3(g) + 3H_3(g) \rightarrow 2NH_3(g)$$

- 1. Consider the reaction in Model 1.
  - st. What are the coefficients for each of the following substances in the reaction?

Draw particle models below to illustrate the reaction in Model 1.



- 2. Consider each situation below as it relates to the reaction in Model 1.
  - a. Calculate the amount of reactants consumed and products made.
  - 6. Record the ratio of N2 to H2 to NH3. Reduce the ratio to the lowest whole numbers possible.

	N <sub>2</sub> Consumed	H <sub>2</sub> Consumed	NH, Produced	Ratio N <sub>3</sub> :H <sub>2</sub> :NH <sub>3</sub> (reduced)
For a single reaction, how many molecules of each substance would be consumed or produced?	1 molecule	3 molecules	2 molecules	1:3:2
If the reaction occurred one hun- dred times, how many molecules would be consumed or produced?	100 molecules	300 molecules	200 molecules	1:3:2
If the reaction occurred 538 times, how many molecules would be consumed or produced?	538 molecules	1614 molecules	1076 molecules	1:3:2

## **Mole Ratios Pogil Answers**

**Kathleen Armour** 

**Mole Ratios Pogil Answers:** 

### Decoding Mole Ratios Pogil Answers: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Mole Ratios Pogil Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers embark on an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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