

Name _____ Date: _____

The Mole Webquest

Use the following website to answer this set of questions:

<http://antoine.frostburg.edu/chem/senese/101/moles/faq/why-use-moles.shtml>

- 1) A mole of anything is how many? (give the number):
- 2) Why is it that different amounts of things can still equal one mole? (think about the weight of a dozen elephants vs a dozen eggs)
- 3) Why do we want to use the concept of moles?
- 4) Once we know the number of moles we can convert to the number of:
_____ or _____ and vice versa.
- 5) How many grams of water are in one mole of water?
- 6) How many molecules of water are in one mole of water?

Use the following website to answer the next set of questions:

<https://chemfiesta.wordpress.com/2015/02/06/mole-calculations-3/>

- 7) Define molar mass.
- 8) What is the mass of one mole of oxygen?
- 9) Find the molar mass of NH_4NO_3 . (Show your work)

Converting between grams, moles, and particles

- 10) Copy the chart:

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- 11) If you are converting between particles and moles, use _____ as conversion factor.

Mole Webquest Key

Barbara Farley Bannister

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Mole Webquest Key:

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McGraw-Hill Staff,2001-03 *Voyages : Französisch für Erwachsene. 3, B1 : Lehr- und Arbeitsbuch : Audio-CD* ,2009

Physical Science George A. Williams,Richard Barnes,Ray Doerhoff,Max Bolen,1979 A physical science text stressing an awareness of the environment with related laboratory activities to lead the student into discovering basic laws and concepts of physics and chemistry Introduction to Chemical Principles Howard Stephen Stoker,1986 **ACT Victory** Cambridge Educational Services, Incorporated,1998

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