

(1) Given this equation: N2 + 3 H2 ---> 2 NH3, write the following molar ratios:

② Given the following equation: 8 H₂ + S₈ --> 8 H₂S, write the following molar ratios:

(3) Answer the following questions for this equation: 2 H₂ + O₂ ---> 2 H₂O

b) Suppose you had 20 moles of
$$H_2$$
 on hand and plenty of O_2 , how many moles of H_2O could you make? 20 moles H_2 2 H_2O = 20 moles H_2O

4) Use this equation: N2 + 3 H2 --> 2 NH3, for the following problems

d) If 0.600 moles of NH3 were produced, how many moles of H2 are required?

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