

ChemThink: Molecular Shapes

Objective:




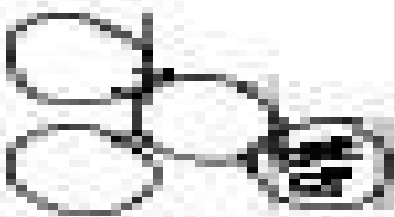

Lesson 8

Electrons

Page

Electron pairs

Understanding that as the number of molecules that have lone pairs on an atom, the electron pairs will repel each other and this will repel and it affect the shape of the molecule.

Number of atoms	Electron pairs	Bond Angles	Shape	Ex. Lewis Dot representation
2		180°	Linear	$H-H$
3		120°	Trigonal Planar	
4		109°	Bent	

Molecular Shapes Answers Chemthink

Lei Shi

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Molecular Shapes Answers Chemthink:

Molecular Shapes Answers Chemthink Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Molecular Shapes Answers Chemthink**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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