

Part 1: Design

10 marks

1. **Identify listed issues:** List 3 issues in the given scenario possible design ideas. Identify three ideas, providing as much detail as possible.
2. **Evaluate and design:** Select one idea and identify a design idea and identify a design idea based on the scenario of each scenario's idea.
3. **Design brief:** Write a design brief with detailed specifications for the final design idea, covering issues such as size, materials, safety, and functionality.
4. **Flow chart:** Write a flow chart to present with the design and construction phase, and each issue is a flowchart to outline the steps to be taken.

Part 2: Planning and Making

10 marks

1. **Working drawings:** List 3 issues in the given scenario possible design ideas for an aspect of the plan, using first angle orthographic projection. Draw that you can a suitable scale, correct line types, and dimensions.
2. **Budget:** Prepare a budget for the "final design" solution, including necessary materials and labour costs. Draw that you budget is suitable and based on real market costs.
3. **Model:** Build a scale model of your final design idea. The model should be constructed using real materials and show intelligent use of materials, design practice safety, when working with tools and materials.

Part 3: Evaluation

10 marks

1. **Evaluation instrument:** Develop an instrument to evaluate the solution. The instrument will be used to assess your team's solution and another team's solution. The will use this during the presentation phase.

Part 4: Communication

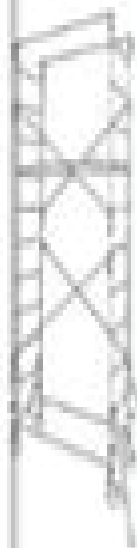
10 marks

1. **Team Presentation:** Teams will present their solution to the "final design" (your teacher and classmates). Each team member must be responsible for presenting a specific aspect of the project. Your presentation should include drawings, plans, the budget, the model, and suitable responses.

Contractors

Design
Phase
1/2

10 marks



Printable

Name:

Class:

Grade 9 CAPS Technology

Mini-PAT

Community Issues – The Contractors

In this Mini-PAT (Practical Assessment Task), you will act as part of a contracting company competing for a contract to design and build a structure that addresses a problem facing a community. The project will focus on identifying a problem, investigating possible solutions, designing a structure, costing the project, and evaluating the final solution. You will be working in a group.

Instructions

Part 1: The Tender Process (including Ethical Practices)

10 marks

1. **Scenario Introduction:** A community lives on a major river's far bank. The river is wide and deep, preventing the people from reaching the city on the opposite side. This means that the people cannot take advantage of the work opportunities, the excellent medical and schools facilities, and the wide range of shops available in the city.

2. **Investigation:** Teams discuss possible structures to solve

ink saving Eco



Mini Pat Grade 9

Annelies Wilder-Smith



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Oxford Textbook of Medical Education Kieran Walsh, 2013-10 Providing a comprehensive and evidence based reference guide for those who have a strong and scholarly interest in medical education the Oxford Textbook of Medical Education contains everything the medical educator needs to know in order to deliver the knowledge skills and behaviour that doctors need The book explicitly states what constitutes best practice and gives an account of the evidence base that corroborates this Describing the theoretical educational principles that lay the foundations of best practice in medical education the book gives readers a through grounding in all aspects of this discipline Contributors to this book come from a variety of different backgrounds disciplines and locations producing a book that is truly original and international The Foundation Programme for Doctors Ferras Alwan, Rohin Francis, Emma Jane Smith, 2018-04-19 This book includes a foreword by Jane Dacre Academic Vice President Royal College of Physicians Professor of Medical Education CHIME Royal Free and University College Medical School University College London Consultant Physician and Rheumatologist The Whittington Hospital London This book prepares you for the often daunting life after medical school detailing everything you need to know about preparing applying and surviving It s the most up to date guide available filling you in on recent important changes to the application system This practical guide is full of valuable tips tables worked examples frequently asked questions further resources and useful addresses Highly recommended for final year medical students and other clinical medical students preparing for their final year this book is also invaluable for junior doctors progressing through the Foundation Programme and all those with teaching and supervisory responsibilities in medical schools and hospitals An invaluable resource for our new generation of doctors It takes readers through the process from application to F2 and beyond It offers useful advice in a useable and readable format It is written by a group of current and past medical students who have lived through and continue to live through the insecurities of the changing medical career structures Its style is informal engaging and easy to absorb so it should be a good distraction for those currently in the run up to their finals exams Good luck to all of you and don t forget Medicine is a wonderful career Jane Dacre in the Foreword A Career in Medicine: Do you have what it takes? second edition Rameen Shakur, 2006-12-19 Have you ever thought about becoming a doctor but weren t sure where to start or what to expect This book provides a comprehensive resource to guide you through the various stages involved in becoming a doctor from deciding whether medicine is right for you to choosing and applying to medical school interview skills student finance plus what *Chain Store Age* ,1962 Journal of the Royal Society of Medicine Royal Society of Medicine (Great Britain), 2009 Includes selected papers from meetings of the Society and of its sections **Manufactured Milk Products Journal** ,1963 **Southern Florist and Nurseryman** ,1964 **Electro Technology Newsletter** Stanley A. Dennis, Leon Irving Thomas, 1955 **The Farmer's Weekly** ,1928 **American Machinist** ,1905 *George Broomhall's Corn Trade News* ,1908 **The Sketch** ,1921 *Shoe and Leather Reporter* ,1929 **Power**

,1901 **The Northwestern Miller** ,1909 **The Motor Boat** Francis P. Prial,1906 **The American Stationer** ,1912
 Chemical Engineering Catalog ,1920 **Farmer's Weekly** ,1928 *The United States Army and Navy Journal and*
Gazette of the Regular and Volunteer Forces ,1921

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) + \text{C}_2\text{H}_2(\text{g}) \rightarrow \text{C}_2\text{H}_4(\text{g}) + \text{O}_2(\text{g})$ $\Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g})$ $\Delta h = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l})$ $\Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l})$ $\Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta h = 285 \text{ kJ/mole}$

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 hess s law example 12.1 determine Δh for the target reaction
 $\text{C}_2\text{H}_4(\text{g}) + \text{O}_2(\text{g}) \rightarrow \text{C}_2\text{H}_4\text{O}(\text{g})$
 given the following information

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 $\text{Mg}(\text{s}) + \frac{1}{2} \text{O}_2(\text{g}) \rightarrow \text{MgO}(\text{s})$ we used

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 cohen 2016 the equations for the reactions in the experiment done are as follows
 $\text{NaOH}(\text{s}) + \text{Na}(\text{aq}) \rightarrow \text{NaOH}(\text{aq})$

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web thus for reaction 9 2 in which 1 mole of H_2SO_4 and 1 mole of $NaOH$ combine to form 2 moles of H_2O $\Delta H = -55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{products} - \Delta H_{reactants}$ is positive if heat is absorbed that is if $\Delta H_{products}$

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