May 2009 TZ1 Question 1

(b) For part of her extended essay investigation into the efficiency of the process, a student reacted a pure sample of a vegetable oil (where R=C₁,H₂₀) with methanol. The raw data recorded for the reaction is below.

Mass of oil	100	1013.0 g
and and the control of the control o	-	200.0 g
Mass of sodium hydroxide	-	3.5 g
Mass of biodiesel produced	1000	811.0 g

(in moles) of the oil and the methanol used, and hence the amount (in moles) of excess methanol.

The relative molecular mass of the oil used by the student is 885.6. Calculate the amount

131

May 2009 TZ2 Question 1

(b) A student reacted some salicylic acid with excess ethanoic anhydride. Impure solid aspirin was obtained by filtering the reaction mixture. Pure aspirin was obtained by recrystallization. The following table shows the data recorded by the student.

Mass of salicylic acid used	$3.15 \pm 0.02 \text{ g}$
Mass of pure aspirin obtained	2.50 ± 0.02 g

- (i) Determine the amount, in mol, of salicylic acid, C_sH_s(OH)COOH, used. [2]
- (ii) Calculate the theoretical yield, in g, of aspirin, C_sH_s(OCOCH_s)COOH. /2/

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Materials Chemistry Bradley D. Fahlman, 2011-03-18 The 2nd edition of Materials Chemistry builds on the strengths that were recognized by a 2008 Textbook Excellence Award from the Text and Academic Authors Association TAA Materials Chemistry addresses inorganic organic and nano based materials from a structure vs property treatment providing a suitable breadth and depth coverage of the rapidly evolving materials field in a concise format The 2nd edition continues to offer innovative coverage and practical perspective throughout e g the opening solid state chemistry chapter uses color illustrations of crystalline unit cells and digital photos of models to clarify their structures This edition features more archetypical unit cells and includes fundamental principles of X ray crystallography and band theory. In addition an ample amorphous solids section has been expanded to include more details regarding zeolite syntheses as well as ceramics classifications and their biomaterial applications. The subsequent metals chapter has been re-organized for clarity and continues to treat the full spectrum of powder metallurgical methods complex phase behaviors of the Fe C system and steels and topics such as corrosion and shape memory properties. The mining processing of metals has also been expanded to include photographs of various processes occurring in an actual steelmaking plant The semiconductor chapter addresses evolution and limitations solutions of modern transistors as well as IC fabrication and photovoltaics Building on the fundamentals presented earlier more details regarding the band structure of semiconductors is now included as well as discussions of GaAs vs Si for microelectronics applications and surface reconstruction nomenclature The emerging field of soft lithographic patterning is now included in this chapter and thin film deposition methodologies are also greatly expanded to now include more fundamental aspects of chemical vapor deposition CVD and atomic layer deposition ALD The polymer and soft materials chapter represents the largest expansion for the 2nd edition This chapter describes all polymeric classes including dendritic polymers as well as important additives such as plasticizers and flame retardants and emerging applications such as molecular magnets and self repairing polymers. This edition now features click chemistry polymerization silicones conductive polymers and biomaterials applications such as biodegradable polymers biomedical devices drug delivery and contact lenses Final chapters on nanomaterials and materials characterization techniques are also carefully surveyed focusing on nomenclature synthetic techniques and applications taken from the latest scientific literature The 2nd edition has been significantly updated to now include nanotoxicity vapor phase growth of 0 D nanostructures and more details regarding synthetic techniques and mechanisms for solution phase growth of various nanomaterials Graphene recognized by the 2010 Nobel Prize in Physics is now also included in this edition Most appropriate for Junior Senior undergraduate students as well as first year graduate students in chemistry physics or engineering fields Materials Chemistry may also serve as a valuable reference to industrial researchers Each chapter concludes with a section that describes important materials applications and an updated list of thought provoking questions. The appendices have also

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- 1. Understanding the eBook May 2009 Chemistry Sl Paper
 - o The Rise of Digital Reading May 2009 Chemistry Sl Paper
 - Advantages of eBooks Over Traditional Books
- 2. Identifying May 2009 Chemistry Sl Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an May 2009 Chemistry Sl Paper
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from May 2009 Chemistry Sl Paper
 - Personalized Recommendations
 - May 2009 Chemistry Sl Paper User Reviews and Ratings
 - May 2009 Chemistry Sl Paper and Bestseller Lists
- 5. Accessing May 2009 Chemistry Sl Paper Free and Paid eBooks
 - May 2009 Chemistry Sl Paper Public Domain eBooks
 - May 2009 Chemistry Sl Paper eBook Subscription Services
 - May 2009 Chemistry Sl Paper Budget-Friendly Options
- 6. Navigating May 2009 Chemistry Sl Paper eBook Formats

- ∘ ePub, PDF, MOBI, and More
- May 2009 Chemistry Sl Paper Compatibility with Devices
- May 2009 Chemistry Sl Paper Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - o Adjustable Fonts and Text Sizes of May 2009 Chemistry Sl Paper
 - Highlighting and Note-Taking May 2009 Chemistry Sl Paper
 - o Interactive Elements May 2009 Chemistry Sl Paper
- 8. Staying Engaged with May 2009 Chemistry Sl Paper
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers May 2009 Chemistry Sl Paper
- 9. Balancing eBooks and Physical Books May 2009 Chemistry Sl Paper
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection May 2009 Chemistry Sl Paper
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine May 2009 Chemistry Sl Paper
 - Setting Reading Goals May 2009 Chemistry Sl Paper
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of May 2009 Chemistry Sl Paper
 - Fact-Checking eBook Content of May 2009 Chemistry Sl Paper
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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