

NACE CP2 – EXAM / Questions & Answers 2023 (NACE Student Manual!)

START OF CHAPTER 1

Corrosion is an electrochemical process involving

- A) flow of protons and electrons
- B) flow of neutrons and ions
- C) flow of protons and ions
- D) flow of electrons and ions
- E) flow of resistance and resistivity

✓ Ans - d

Metal loss or corrosion occurs at the

- A) cathode
- B) anode
- C) electronic path
- D) metallic path

✓ Ans - b

What are the four parts of the corrosion cell

- A) anode, resistor, electrolyte, metallic path
- B) anode, cathode, electron, electronic path
- C) anode, cathode, electrolyte, metallic path

✓ Ans - c

True or False (t or f):

Ions exist only in the electrolyte?

✓ Ans - t

Electrons flow in a corrosion cell by which of the following paths?

- A) anode to cathode via electrolyte
- B) cathode to anode via electronic path
- C) anode to cathode via electronic path

✓ Ans - c

At the cathodic surface in a corrosion cell, the electrons from the ____ meet ____ ions from the solution.

- A) cathode, hydrogen
- B) cathode, OH⁻
- C) anode, hydrogen
- D) anode, OH⁻

✓ Ans - c

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G Psacharopoulos



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