

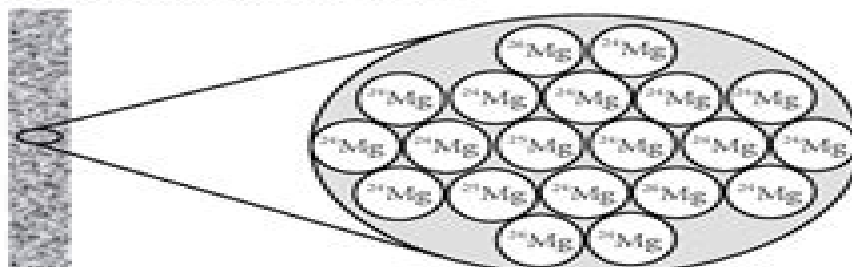
Average Atomic Mass

How are the masses on the periodic table determined?

Why?

Most elements have more than one naturally occurring isotope. As you learned previously, the atoms of those isotopes have the same atomic number (number of protons), making them belong to the same element, but they have different mass numbers (total number of protons and neutrons) giving them different atomic masses. So which mass is put on the periodic table for each element? Is it the most common isotope's mass? The heaviest mass? This activity will help answer that question.

Model 1 – A Strip of Magnesium Metal



1. Write in the atomic number for each Mg atom in Model 1.

All are 12

2. What are the mass numbers of the naturally occurring isotopes of magnesium shown in Model 1?

24, 25, 26

3. Do all of the atoms of magnesium in Model 1 have the same atomic mass? Explain.

No there are 3 different mass numbers which mean there are 3 different masses

4. For the sample of 20 atoms of magnesium shown in Model 1, draw a table indicating the mass numbers of the three isotopes and the number of atoms of each isotope present.

Mass #	# of atoms
24	16
25	2
26	2

5. Which isotope of magnesium is the most common in Model 1?

Magnesium 24 is the most common

6. Based on Model 1 and the table you created in Question 4, for every 10 atoms of magnesium, approximately how many atoms of each isotope will be found?

Mg 24=8 Mg 25=1 Mg 26=1

Isotopes And Average Atomic Mass Answer Key

Neil D. Jespersen, Alison Hyslop



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Regents Chemistry--Physical Setting Power Pack Revised Edition Albert S. Tarendash, 2021-01-05 Barron's two book Regents Chemistry Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Chemistry Regents exam This edition includes Regents Exams and Answers Chemistry Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day Let's Review Regents Chemistry Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key Let's Review Regents: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert S. Tarendash, 2021-01-05 Barron's Let's Review Regents Chemistry gives students the step by step review and practice they need to prepare for the Regents Chemistry Physical Setting exam This updated edition is an ideal companion to high school textbooks and covers all Chemistry topics prescribed by the New York State Board of Regents Let's Review Regents Chemistry covers all high school level Chemistry topics and includes Extensive review of all topics on the test Extra practice questions with answers A detailed introduction to the Regents Chemistry course and exam One actual recently released Regents Chemistry exam with an answer key *Ebook: Introductory Chemistry: An Atoms First Approach* Burdge, 2016-04-16 Ebook Introductory Chemistry An Atoms First Approach Numerical Chemistry for Competitions Anu Sharma, 2009 An ideal book for the students of XI and XII CBSE ISC and the State Boards who are using Core Curriculum and also useful for the students preparing for various Engineering Medical Entrance Examinations **Oswaal NCERT Textbook Solution Class 11 | Physics | Chemistry | Biology | Set of 3 Books | For Latest Exam** Oswaal Editorial Board, 2024-03-30 Description of the Product Updated for 2024 25 The books are 100% updated for the academic year 2024 25 adhering strictly to the latest NCERT guidelines Comprehensive Coverage We cover all concepts and topics outlined in the most recent NCERT textbooks Visual Learning Aids Explore theoretical concepts and concept videos that offer a brief description of the topic and help visualize complex concepts Effective Revision Tools Benefit from crisp Revision Notes Mind Maps and Mnemonics designed to facilitate efficient and effective review Complete Question Coverage All questions from the NCERT textbooks are covered in our solutions providing a thorough grasp of the subject matter *Foundation Course for NEET (Part 2): Chemistry Class 9* Lakhmir Singh & Manjit Kaur, Our NEET Foundation series is sharply focused for the NEET aspirants Most of the students make a career choice in the middle school and therefore choose their stream informally in secondary and formally in senior secondary schooling accordingly If you have decided to make a career in the medical profession you need not look any further Adopt this series for Class 9 and 10 today **Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-12-03 Chemistry The

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clear and logical explanations of chemical concepts as well as the right mix of general chemistry organic chemistry and biochemistry An emphasis on real world topics lets readers clearly see how the chemistry will apply to their career

General Chemistry for Engineers James O. Glanville,2002 Emphasizing problem solving and engineering approximation this chemistry book provides engineers with an understanding of the entities atoms molecules and ions that are relevant to their lives and professional careers Throughout the book internet key word searching and graphing exercises take advantage of users existing computer skills and encourages them to acquire new ones in designing preparing and interpreting graphs Chapter topics cover atoms elements and measurements nuclides molecules and ions chemical reaction and stoichiometry gases quantum mechanics and the periodic table chemical bonding and chemical structure chemical energy and the first law of thermodynamics the second law of thermodynamics and chemical equilibrium gas and solution equilibria liquids and their mixtures solids phase diagrams and solutions the periodic table and redox chemistry electrochemistry and rate processes For engineers preparing for the professional certification exam Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction **General Chemistry** Brady,1990-01-02 *Merrill Chemistry* Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *World of Chemistry* Melvin D. Joesten,2004

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