

Name: \_\_\_\_\_

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# Mixtures and Solutions Study

## 1. What is a mixture?

Answer: A mixture is a combination of two or more substances that are physically combined but not chemically bonded. The components of a mixture retain their individual properties and can be separated by physical means (e.g., filtration, evaporation).

## 2. What is a solution?

Answer: A solution is a type of mixture where one substance (the solute) is completely dissolved in another substance (the solvent). Solutions are homogeneous, meaning the composition is uniform throughout.

## 3. Types of Mixtures

Homogeneous Mixture: The composition is uniform throughout. (Example: Saltwater)

Heterogeneous Mixture: The composition is not uniform; different parts of the mixture are visible. (Example: Salad or a trail mix)

## 4. Examples of Mixtures

Homogeneous Mixture: Sugar dissolved in water, air (a mixture of gases).

Heterogeneous Mixture: Sand and iron filings, salad, chocolate chip cookies.

## 5. What are solutes and solvents?

Solute: The substance that is dissolved in a solution.

Solvent: The substance that dissolves the solute.

## 6. What is a saturated solution?

Answer: A saturated solution contains the maximum amount of solute that can dissolve at a given temperature. Any additional solute will not dissolve and will settle at the bottom.

## 7. What is solubility?

Answer: Solubility is the ability of a solute to dissolve in a solvent at a specific temperature and pressure. It varies for different substances.

## 8. What is a suspension?

Answer: A suspension is a heterogeneous mixture where the particles are large enough to settle out over time (e.g., muddy water).

## 9. What is a colloid?

Answer: A colloid is a type of mixture where the particles are dispersed throughout but are not large enough to settle out (e.g., milk, fog). Colloids scatter light, known as the Tyndall effect.

## 10. What is concentration?

Answer: Concentration refers to the amount of solute dissolved in a certain amount of solvent or solution. A solution can be dilute (low concentration of solute) or concentrated (high concentration of solute).

# Mixtures And Solutions Study Guide Answers

**Terry C. Jones**



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