PONNE

2.	Sodium thiosulfate solution,	Na2S2O3(aq),	and	hydrochloric	acid,	HCl (aq),	react to	produce
	solid sulfur as in the equation	below.					A COLUMN TO A COLU	

$$S_2O_3^{2-}(aq) + 2H^*(aq) \rightarrow S(s) + SO_2(g) + H_2O(l)$$

The following results to determine the initial rate were obtained:

MI ALAK THEM WHIPE HAVE TZENK

Experiment	[S ₂ O ₃ ² (aq)] / moldm ²	[H'(aq)] / mol dm-3	Initial rate / moldm's
	0.200	2.00	0.036
2	0.200	1.00	0.036
3	0.100	1.00	0.018

(a)	Deduce, with a reason, the order of reaction with respect to each reactant.	[2]
(b)	State the rate expression for this reaction.	[1]
138		
No.		
(c)	Determine the value of the rate constant, k, and state its units.	[2]
182		
1337	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	
100		300

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KJ Lindholm-Leary

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