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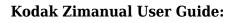
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# **Kodak Zimanual User Guide**

**Eastman Kodak Company of New York** 



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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic eg 1 h cl aq n aoh aq n acl aq h 20 l for which  $\Delta h$  57kj mol 1 eg 2 h n o3 aq koh aq kn o3 aq h 20 l for which  $\Delta h$  57kj mol 1 3 1 flashcards by dapz ogunfuye brainscape - Apr 17 2022

web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example hcl aq naoh aq nacl aq h2o aq you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts