

ACIDS AND BASES QUIZ

1. Which of the following indicates an acidic solution?

- A) pH = 7.5 B) pH = 7 C) pH = 6.5
D) more than one correct response E) no correct response

2. Which of the following equations represents an acid-base neutralization reaction?

- A) $\text{H}_2\text{SO}_4 + \text{Zn} \rightarrow \text{ZnSO}_4 + \text{H}_2$ B) $\text{HNO}_3 + \text{KOH} \rightarrow \text{KNO}_3 + \text{H}_2\text{O}$
C) $\text{Ba}(\text{OH})_2 + \text{Na}_2\text{SO}_4 \rightarrow \text{BaSO}_4 + 2\text{NaOH}$ D) more than one correct response
E) none of the above

3. In which of the following pairs of substances are both members of the pair salts?

- A) NaCl and NaOH B) KNO_3 and K_2SO_4 C) H_3PO_4 and $(\text{NH}_4)_3\text{PO}_4$
D) more than one correct response E) none of the above

4. If the pH of a solution increases from 4.0 to 6.0, the hydronium ion concentration

- A) increases by a factor of 1.5. B) increases by a factor of 100.
C) decreases by a factor of 1.5. D) decreases by a factor of 100.

5. Which of the following statements concerning Arrhenius acids and Arrhenius bases is incorrect?

- A) In the pure state, Arrhenius acids are covalent compounds.
B) In the pure state, Arrhenius bases are ionic compounds.
C) Dissociation is the process by which Arrhenius acids produce H^+ ions in solution.
D) Arrhenius bases are also called oxides.

6. Highly basic solutions are:

- A) safe to handle B) sweet tasting
C) corrosive and reactive like highly acidic solutions D) corrosive but not reactive

7. When dissolved in water, sodium hydroxide (NaOH):

- A) turns phenolphthalein colourless. B) produces a high concentration of hydronium ions.
C) releases O^{2-} ions. D) produces a high concentration of hydroxide ions.
E) reacts with carbonates to produce a salt and carbon dioxide gas.

8. Bases taste:

- A) Salty B) Bitter C) Sweet D) Sour

9. The characteristic ion of acids is:

- A) carbonate ion CO_3^{2-} B) phosphate ion PO_4^{3-} C) hydrogen ion (H^+) D) hydroxide ion (OH^-)

10. Which of the following pH values indicates the strongest base?

- A) 4 B) 7 C) 11 D) 1

11. A positive test for a base occurs when:

Multiple Choice On Acids And Bases General Chem

David R. Klein



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Fundamentals of General, Organic, and Biological Chemistry Daniel Sullivan (Prof. of Chemistry.), 1999

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Chem 5e Irm Steven S. Zumdahl, 2000-02

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The Success Manual for General Chemistry Elizabeth Kean, Catherine Middlecamp, 1986

Study Guide, Chemical Principles, Fifth Edition, Zumdahl Paul B. Kelter, Steven S. Zumdahl, 2004-04

How to Survive (and Even Excel In) General Chemistry Elizabeth Kean, Catherine Middlecamp, 1994 A different kind of book about chemistry which teaches readers the process of learning chemistry not the topic itself Proving a valuable supplement to any introductory text this guide offers inside information to help make chemistry less stressful even enjoyable Includes exercises and sections for self assessment

Study Guide and Solutions Manual for McMurry's Organic Chemistry, Fifth Edition Susan McMurry, 2000 Provides

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Saunders Interactive General Chemistry CD-ROM Mary L Kotz, John C. Kotz, William J. Vining, Justin T. Fermann, 1998-11

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1. Understanding the eBook Multiple Choice On Acids And Bases General Chem
 - The Rise of Digital Reading Multiple Choice On Acids And Bases General Chem
 - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice On Acids And Bases General Chem
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Multiple Choice On Acids And Bases General Chem
 - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice On Acids And Bases General Chem
 - Personalized Recommendations
 - Multiple Choice On Acids And Bases General Chem User Reviews and Ratings
 - Multiple Choice On Acids And Bases General Chem and Bestseller Lists
5. Accessing Multiple Choice On Acids And Bases General Chem Free and Paid eBooks
 - Multiple Choice On Acids And Bases General Chem Public Domain eBooks
 - Multiple Choice On Acids And Bases General Chem eBook Subscription Services
 - Multiple Choice On Acids And Bases General Chem Budget-Friendly Options

6. Navigating Multiple Choice On Acids And Bases General Chem eBook Formats
 - ePub, PDF, MOBI, and More
 - Multiple Choice On Acids And Bases General Chem Compatibility with Devices
 - Multiple Choice On Acids And Bases General Chem Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Multiple Choice On Acids And Bases General Chem
 - Highlighting and Note-Taking Multiple Choice On Acids And Bases General Chem
 - Interactive Elements Multiple Choice On Acids And Bases General Chem
8. Staying Engaged with Multiple Choice On Acids And Bases General Chem
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Multiple Choice On Acids And Bases General Chem
9. Balancing eBooks and Physical Books Multiple Choice On Acids And Bases General Chem
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Multiple Choice On Acids And Bases General Chem
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Multiple Choice On Acids And Bases General Chem
 - Setting Reading Goals Multiple Choice On Acids And Bases General Chem
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Multiple Choice On Acids And Bases General Chem
 - Fact-Checking eBook Content of Multiple Choice On Acids And Bases General Chem
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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