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## Introduction

Chromatography is a technique that is used to separate different solutions or mixtures into their different components by running the mixture through a separation column. Different types of chromatography are based on varying physical properties of the components, such as polarity, charged residues, or differences in size. Size-exclusion and ion-exchange chromatography

were performed in this laboratory to separate four known proteins from a protein mixture. The experiments were conducted to understand the theory behind chromatography by applying the classroom knowledge in a practical setting. And, the spectrographs of the elution were taken and analysed to quantify the strength of separation. The results from a protein solution containing mixtures of four proteins were plotted and evaluated to test the accuracy of the conducted chromatographies.

## 2 Body

### 2.1 Theory

Size-exclusion (SE) chromatography separates components based on their sizes. A porous, hydrophilic support is used to reduce the potential of intermolecular bonding. This support is usually cross-linked between dextrans and cellulose<sup>[1]</sup> and is transported throughout the column as means of inserting a flow of a suitable mobile phase. As the protein mixture travels down the column, the smaller components could enter the pores of the support. As a result, smaller proteins are maintained in the stationary phase for longer and elute out of the column more slowly than larger molecules. Accordingly, the first peaks of the chromatogram show the largest proteins in the solutions.

Ion-exchange (IE) chromatography takes advantage of the difference of charges between the regions of a molecule and a charged-group on the beaded resin (stationary phase)<sup>[1]</sup>. Each protein has an iso-electric point (pI). This is the pH at which the overall charge on the surface of the protein is 0<sup>[2]</sup>. When the protein is on the mobile phase that has a pH greater than the isoelectric point, it would be negatively charged. Conversely, if the pH of the solution is lower than the isoelectric point of the protein, its surface is negatively charged<sup>[1]</sup>. As the mobile phase, along with the analyte, flows through the column, the components that are most reactive with the resin are retained longer, leaving the column lastly.

A spectrometer measures absorbances at different wavelengths and plots them against the times or fraction number.

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# Manual Document Editing

**Ivan Flores**



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5 Adding Comments and Annotation 6 The Comment Pane 7 Advanced Comments List Option Commands 8 Enabling  
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### **Table of Contents Manual Document Editing**

1. Understanding the eBook Manual Document Editing
  - The Rise of Digital Reading Manual Document Editing
  - Advantages of eBooks Over Traditional Books
2. Identifying Manual Document Editing
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Manual Document Editing
  - User-Friendly Interface
4. Exploring eBook Recommendations from Manual Document Editing
  - Personalized Recommendations
  - Manual Document Editing User Reviews and Ratings
  - Manual Document Editing and Bestseller Lists

5. Accessing Manual Document Editing Free and Paid eBooks
  - Manual Document Editing Public Domain eBooks
  - Manual Document Editing eBook Subscription Services
  - Manual Document Editing Budget-Friendly Options
6. Navigating Manual Document Editing eBook Formats
  - ePub, PDF, MOBI, and More
  - Manual Document Editing Compatibility with Devices
  - Manual Document Editing Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Manual Document Editing
  - Highlighting and Note-Taking Manual Document Editing
  - Interactive Elements Manual Document Editing
8. Staying Engaged with Manual Document Editing
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Manual Document Editing
9. Balancing eBooks and Physical Books Manual Document Editing
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Manual Document Editing
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Manual Document Editing
  - Setting Reading Goals Manual Document Editing
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Manual Document Editing
  - Fact-Checking eBook Content of Manual Document Editing
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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