

Chapter 5 McGraw Hill

1. A physical change may involve _____. (Check all that apply.)
 - a change in volume.
 - a change in phase.
 - a change in shape.
2. Which of the following are included in a complete balanced chemical equation? (Select all that apply.)
 - coefficients
 - reactants
 - phases
 - products
 - reaction arrow
3. Which of the following should you do when balancing a chemical equation? (Select all that apply.)
 - Balance the equation with coefficients one element at a time.
 - Check your solution to verify that the same number of each type of atom is present on both sides of the reaction.
4. Which statements about decomposition reactions are correct? Select all that apply.
 - The reactant in a decomposition reaction is always a compound.
 - A decomposition reaction begins with a single reactant.
5. In a _____ reaction, two reactants exchange components to form two new products.
double replacement
6. The boiling of water is an example of a _____ change.
physical
7. Reduction is _____.
the gain of electrons by an atom
8. Which of the following should you do when balancing a chemical equation? (Select all that apply.)
 - Check your solution to verify that the same number of each type of atom is present on both sides of the reaction.
 - Balance the equation with coefficients one element at a time.
9. Identify the species that are oxidized and reduced in the reaction below.
10. $\text{CH}_4\text{O} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
CH₄O is oxidized and O₂ is reduced.

Mcgraw Hill Chemistry Answer Key Ch 5

Sabine Zange



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