INORGANIC CHEMISTRY

451.	Nitrogen can be obtained from air by removing			
	(a) Oxygen	(b) Hydrogen	(c) Carbon dioxide	(d) Both (a) and (c)
452.	Which of the follow (a) Be ₂ C	wing carbide is a methani (b) CaC ₂	ide? (c) Al ₂ C ₆	(d) Mg ₂ C ₃
453.	Caro's acid is			
	(a) H ₂ SO,	(b) H,S,O,	(c) H,SO,	(d) H ₂ S ₂ O ₀
454.	Inorganic benzene is			
	(a) B, H, N,	(b) BH ₃ NN ₃	(c) B ₁ H ₄ N,	(d) H ₃ B ₃ N ₄
455.	For the reaction, P(White) — P(Red); ΔH = -18.4 kJ. It follows that (a) Red P is readily formed from white P (b) White P is readily formed from red P (c) White P can not be converted to red P (d) White P can be converted into red P and red P is more stable			
456	The state of the s	annot be obtained by hea (b) Pb(NO ₁) ₀	The state of the s	
45.7.	(a) KNO ₃ Reactivity of bone	cole is greater than that of	(e) Cu(NO ₃) ₂	(d) AgNO ₃
	(a) Borazole is a n (b) Borazole is a p	on-polar compound olar compound electron deficient compo		
458.	When orthophosphoric acid is heated to 600° C, the product formed is			
	(a) Phosphine, PH,		(b) Phosphorus pentoxide, P ₂ O ₅	
	(c) Phosphorus acid, H,PO, (d) Metaphosphoric acid, HPO,			ed, MPO,
459.	$P_4 + 3NaOH + 3H_2O \rightarrow A + 3NaH_2PO_2$ here, 'A' is			
V	(a) NH,	(b) PH ₃	(c) H,PO,	(d) H ₁ PO ₁
460.	SO ₃ in the solid ph	ase exists as		The second secon
	(a) linear chain or cyclic trimer		(b) planar triongular unit	
NUMBER OF	(c) square planar unit		(d) none of these	
461.	Sulphur does not exist as S ₂ molecule because			
"	(a) It is less electronegative (c) It has ability to exhibiting catenation state		 (b) Does not form pπ-pπ bond (d) Has tendency to show variable oxidation 	
46.2.		wing has S-S BOND?		
463	(a) H ₂ S ₂ O ₆ Which of the follow	(b) H ₂ S ₂ O ₇ wing is correct order of si	(c) H ₂ S ₂ O ₈	(d) Mustard gas
40.5.	(a) Pb2 >Pb4	(b) Pb ⁴ >Pb ²	(c) Si ² >Si ⁴	(d) Sn ⁴ > Sn ²
464.	Quartz is a crystalline variety of			
	(a) Si	(b) SiO ₂	(c) NacSiO ₂	(d) SiC
465.	The equivalent weight of phosphoric acid in the reaction NaOH + H ₃ PO ₄ → NaH ₂ PO ₄ + H ₂ O			
	(a) 25	(b) 49	(c) 59	(d) 98

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