

(iii) Suggest why the mean bond enthalpy of a C=C bond is less than twice the mean bond enthalpy of a C—C bond.

(1)

(iv) Use the mean bond enthalpy data in the table, and the equation given below, to calculate a value for the standard enthalpy change of combustion of propene.

(3)

Bond	Mean bond enthalpy / kJ mol <sup>-1</sup>
C=C	612
C—C	347
C—H	413
O=O	498
C=O	805
O—H	464



Answer = \_\_\_\_\_ kJ mol<sup>-1</sup>



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**H Kauffman**



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**Table of Contents January 2015 Ial Paper C**

1. Understanding the eBook January 2015 Ial Paper C
  - The Rise of Digital Reading January 2015 Ial Paper C
  - Advantages of eBooks Over Traditional Books
2. Identifying January 2015 Ial Paper C
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an January 2015 Ial Paper C
  - User-Friendly Interface
4. Exploring eBook Recommendations from January 2015 Ial Paper C
  - Personalized Recommendations
  - January 2015 Ial Paper C User Reviews and Ratings
  - January 2015 Ial Paper C and Bestseller Lists
5. Accessing January 2015 Ial Paper C Free and Paid eBooks
  - January 2015 Ial Paper C Public Domain eBooks
  - January 2015 Ial Paper C eBook Subscription Services
  - January 2015 Ial Paper C Budget-Friendly Options
6. Navigating January 2015 Ial Paper C eBook Formats
  - ePub, PDF, MOBI, and More
  - January 2015 Ial Paper C Compatibility with Devices
  - January 2015 Ial Paper C Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of January 2015 Ial Paper C
  - Highlighting and Note-Taking January 2015 Ial Paper C
  - Interactive Elements January 2015 Ial Paper C
8. Staying Engaged with January 2015 Ial Paper C

- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers January 2015 Ial Paper C
9. Balancing eBooks and Physical Books January 2015 Ial Paper C
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection January 2015 Ial Paper C
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine January 2015 Ial Paper C
- Setting Reading Goals January 2015 Ial Paper C
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of January 2015 Ial Paper C
- Fact-Checking eBook Content of January 2015 Ial Paper C
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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