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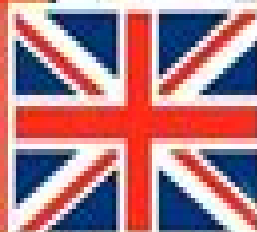
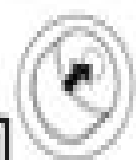


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Manual De Servicio Chevy

**Equipo editorial de Electrónica y
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Manual De Servicio Chevy:

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web exothermic reactions are chemical reactions that produce heat in section ii b of chapter 3 we defined the heat flow q as negative when heat flows from the system to the surroundings thus exothermic reactions have a negative Δh of reaction the word exothermic is derived from exo the greek word for outside and therme the greek word

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web in the context of a chemical reaction the term neutralization is used for a reaction between an acid and a base or alkali historically this reaction was represented as acid base alkali salt water for example $\text{hcl naoh nacl h}_2\text{o}$

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web during this reaction the h^+ ions from the acid and the oh^- ions from the alkali react together to form water as this happens energy is released which is detected by the thermometer

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web reacting two solutions eg acid and alkali place the polystyrene cup inside the glass beaker to make it more stable measure an appropriate volume of each liquid eg 25 cm³ place one of the

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web for reactions involving acetic acid or ammonia the measured enthalpy change of neutralization is a few kJ less exothermic than with strong acids and bases for example one source which gives the enthalpy change of neutralization of sodium hydroxide solution with hcl as 57.9 kJ mol⁻¹

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web sep 20 2014 i will illustrate this with some examples when an acid is neutralised by an alkali the reaction is exothermic

eg 1 $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o l}$ for which $\Delta h = 57\text{ kJ mol}^{-1}$ eg 2 $\text{h}_2\text{so}_4 \text{ aq} + \text{koh aq} \rightarrow \text{ks}_2\text{so}_4 \text{ aq} + \text{h}_2\text{o l}$ for which $\Delta h = 57\text{ kJ mol}^{-1}$

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web the enthalpy change of neutralisation is defined as the enthalpy change when solutions of an acid and alkali react together to produce one mole of water for example $\text{hcl aq} + \text{naoh aq} \rightarrow \text{nacl aq} + \text{h}_2\text{o aq}$ you can see that one mole of water is produced when one mole of hcl or one mole of naoh reacts

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web most ceremonies include a flag or opening ceremony reciting of the girl scout promise and law crossing of a bridge the girl scout handshake presentation of certificates patches and other awards an ending ceremony other popular additions include doing the friendship squeeze singing a girl scout song serving refreshments

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web 1 daisy to brownie 2 brownie to junior 3 junior to cadette 4 cadette to senior 5 senior to ambassador 6 ambassador to adult bridging awards most girl scouts choose to earn the bridging award for their level earning the award offers a chance to look back on what they ve accomplished while looking to the future

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