

## Ch 13 Study Guide (Chemical Bonds)

1. Fill in the table below with the correct values:

Element Name	Ion Symbol	# Protons	# Electrons	Charge	Ion Type
Fluorine	F <sup>-</sup>	9	10	-1	Anion
Iodine	I <sup>-</sup>	53	54	-1	Anion
Sulfur	S <sup>2-</sup>	16	18	-2	Anion
Potassium	K <sup>+</sup>	19	18	+1	Cation
Calcium	Ca <sup>2+</sup>	20	18	+2	Cation
Bromine	Br <sup>-</sup>	35	36	-1	Anion
Strontium	Sr <sup>2+</sup>	38	36	+2	Cation
Oxygen	O <sup>2-</sup>	8	10	-2	Anion
Magnesium	Mg <sup>2+</sup>	12	10	+2	Cation
Aluminum	Al <sup>3+</sup>	13	10	+3	Cation
Selenium	Se <sup>2-</sup>	34	36	-2	Anion
Lithium	Li <sup>+</sup>	3	2	+1	Cation
Rubidium	Rb <sup>+</sup>	37	36	+1	Cation
Chlorine	Cl <sup>-</sup>	17	18	-1	Anion

2. An ionic bond is formed between what two types of atoms?

Metal & nonmetal

3. A covalent bond is formed between what two types of atoms?

Nonmetal & Nonmetal

4. Draw the bond(s) that will form between Mg and O.



5. What is the formula for the compound formed in question #4?



# Ions Answer Key

**Mary Millhollon, Richard H. Langley**



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## **Table of Contents Ions Answer Key**

1. Understanding the eBook Ions Answer Key
  - The Rise of Digital Reading Ions Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Ions Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ions Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ions Answer Key
  - Personalized Recommendations
  - Ions Answer Key User Reviews and Ratings
  - Ions Answer Key and Bestseller Lists
5. Accessing Ions Answer Key Free and Paid eBooks
  - Ions Answer Key Public Domain eBooks
  - Ions Answer Key eBook Subscription Services
  - Ions Answer Key Budget-Friendly Options

6. Navigating Ions Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Ions Answer Key Compatibility with Devices
  - Ions Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Ions Answer Key
  - Highlighting and Note-Taking Ions Answer Key
  - Interactive Elements Ions Answer Key
8. Staying Engaged with Ions Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Ions Answer Key
9. Balancing eBooks and Physical Books Ions Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Ions Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Ions Answer Key
  - Setting Reading Goals Ions Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Ions Answer Key
  - Fact-Checking eBook Content of Ions Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements



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web right triangles pythagorean theorem sines sin cosines cos tangents tan now suppose we know the hypotenuse and one side but have to find the other for example if 119 and 169 then  $119^2 + 28561 = 14161$   $14400$  and the square root of  $14400$  is 120 so 120 we might only know one side but we also know an

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