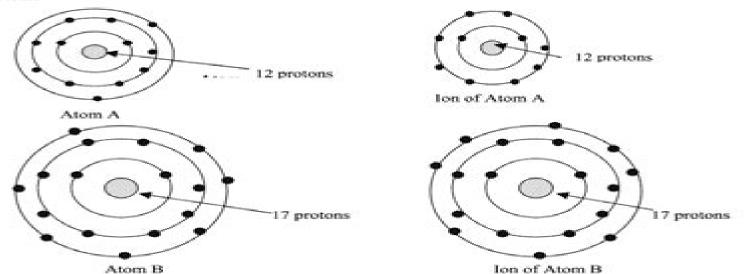
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Figure 1: Below are four Bohr diagrams of atoms and ions. The two diagrams on the left are atoms; the two on the right are ions.



Critical Thinking Questions

- 1. How do you know that both Atom A and Atom B are neutral (have a charge of zero).
- 2. What elements are Atom A and Atom B?
- 3. Given the above diagrams, explain how an atom becomes an ion?
- 4. What is the charge on Ion A? What is the charge on Ion B? How do you know?
- Write the electron configuration (ex: 1s²2s²2p⁶....) for each ion and atom shown in the Bohr diagrams.

Atom A: _____

Ion A: _____

Atom B: _____ Ion B: ____

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Rudolph Max Goepp

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