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web the first two equations have been balanced for you alpha particle α 2 4 beta particle β gamma ray γ proton p neutron n 0 positron β^+ 0 1 30 1 65 1 2 63 2 15 0 30 0 65 30 9 29 2 2 12 6 15 32 30

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web identify the missing particle in the following nuclear reaction $1\ 0\ \text{n} + {}^{235}_{92}\text{u} \rightarrow {}^{140}_{54}\text{x} + {}^{94}_{38}\text{sr} + 2\ 1\ 0\ \text{n}$ ${}^{137}_{52}\text{te} \rightarrow {}^{42}_{2}\text{he} + {}^{97}_{38}\text{sr} + {}^{96}_{40}\text{zr} + 97\ 40\ \text{zr}$

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